

Accepted Manuscript

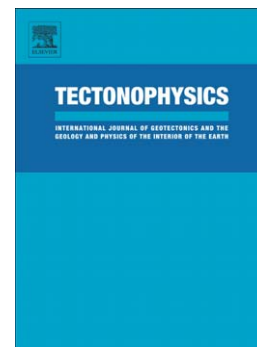
Refertilization process in the Patagonian subcontinental lithospheric mantle of Estancia Sol de Mayo (Argentina)

Massimiliano Melchiorre, Massimo Coltorti, Michel Gregoire, Mathieu Benoit

PII: S0040-1951(15)00132-8
DOI: doi: [10.1016/j.tecto.2015.02.015](https://doi.org/10.1016/j.tecto.2015.02.015)
Reference: TECTO 126554

To appear in: *Tectonophysics*

Received date: 28 May 2014
Revised date: 28 January 2015
Accepted date: 16 February 2015



Please cite this article as: Melchiorre, Massimiliano, Coltorti, Massimo, Gregoire, Michel, Benoit, Mathieu, Refertilization process in the Patagonian subcontinental lithospheric mantle of Estancia Sol de Mayo (Argentina), *Tectonophysics* (2015), doi: [10.1016/j.tecto.2015.02.015](https://doi.org/10.1016/j.tecto.2015.02.015)

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

1 **Refertilization process in the Patagonian subcontinental lithospheric mantle of Estancia**
2 **Sol de Mayo (Argentina)**

3

4 Massimiliano Melchiorre^{1,2,*}, Massimo Coltorti¹, Michel Gregoire³, Mathieu Benoit³

5

6 (1) Department of Physics and Earth Science, Ferrara University, via Saragat 1, 44123 Ferrara, Italy.

7 (2) Institute of Earth Sciences Jaume Almera – CSIC, C. Luís Solé i Sabarís s/n, 08034 Barcelona,
8 Spain.

9 (3) Géosciences Environment Toulouse, CNRS-IRD-Toulouse University, Midi-Pyrénées
10 Observatory, 14 Av. E. Belin, 31400 Toulouse, France.

11

12 * Corresponding author: Massimiliano Melchiorre, , Institute of Earth Science Jaume Almera – CSIC,
13 Barcelona, Spain. *E-mail address:* mmelchiorre@ictja.csic.es

14

15

16 **Abstract**

17

18 Anhydrous mantle xenoliths equilibrated at 1003-1040°C from Estancia Sol de Mayo (ESM, Central
19 Patagonia, Argentina) and entrained in post-plateau alkaline lavas belonging to Meseta Lago Buenos
20 Aires have been investigated aiming at reconstructing the depletion and enrichment processes that
21 affected this portion of the Patagonia lithospheric mantle. Xenoliths are characterized by a coarse-
22 grained protogranular texture and are devoid of evident modal metasomatism. They show two texturally
23 different clinopyroxenes: protogranular (cpx1) and texturally related to spinel (cpx2). Three different
24 types of orthopyroxenes are also recognized: large protogranular crystals with exsolution lamellae
25 (opx1); small clean and undeformed grains without exsolution lamellae (opx2) and small grains
26 arranged in vein (opx3). Major element composition of clinopyroxenes and orthopyroxenes highlights
27 two different trends characterized by i) high Al₂O₃ content at almost constant mg# and ii) a slight
28 increase in Al₂O₃ content with decreasing mg. Clinopyroxenes are enriched in LREE and are
29 characterized by prominent to slightly negative Nb, Zr and Ti anomalies. No geochemical differences

30 are observed between cpx1 and cpx2, whilst a discrimination can be observed between opx1 and opx2
31 (LREE-depleted; prominent to slightly negative Ti and Zr anomalies) and opx3 (prominent positive Zr
32 anomaly). Partial melting modelling using both major and trace elements indicates a melting degree
33 between ~ 5% and ~ 13% (up to ~ 23% according to major element modelling) for lherzolites and
34 between ~ 20% and ~ 30% for harzburgites (down to ~ 5% according to trace element modelling).
35 La/Yb and Al₂O₃, as well as Sr and Al₂O₃ negative correlation in clinopyroxenes point to a
36 refertilization event affecting this lithospheric mantle. The agent was most probably a transitional
37 alkaline/subalkaline melt, as indicated by the presence of orthopyroxene in vein and the similar
38 geochemical features of ESM clinopyroxenes and those from Northern Patagonia pyroxenites which are
39 derived from transitional alkaline/subalkaline lavas.

40

41 **Key words:** Patagonia, mantle xenoliths, refertilization, T-P-/O₂ conditions

42

43

44 **1. Introduction**

45 Xenoliths of sub-arc mantle entrained in arc magmas are rarer than those from intra-plate settings, i.e.
46 from oceanic hotspots and continental rift zones (*Nixon, 1987*). Thus a paucity of xenolith-based, direct
47 petrological information of the mantle wedge exists and a systematic investigation of the rare
48 occurrences of these xenoliths needs to be carried out. Within subduction settings two groups of
49 xenoliths can be distinguished taking into account the composition of the host magma, i.e. alkaline and
50 calc-alkaline s.l. lavas. The type locality of xenoliths entrained in alkaline basalts in back arc zones is
51 Patagonian, but several occurrences have been found also in the Mediterranean area [Tallante, *Bianchini*
52 *et al., 2011*; Pannonian Basin, including Styrian Basin (Kapfenstein, *Kurat et al., 1980*; *Coltorti et al.,*
53 *2007*, Bakoni–Balaton Highlands, *Bali et al., 2008*; *Hidas et al., 2010*; *Berkesi et al., 2012*)]. Xenoliths
54 entrained in calc-alkaline basalts s.l. are those belonging to the Japan arc (*Takahashi 1978*; *Aoki 1987*;
55 *Arai et al. 1998*), the Kamchatka arc (*Koloskov and Khotin, 1978*; *Kepezhinskas et al., 1995*; *Arai et*

56 *al.*, 2003; *Widom et al.*, 2003; *Saha et al.*, 2005; *Weyer & Ionov*, 2007; *Ionov and Seitz*, 2008), and the
57 Tabar–Lihir–Tanga–Feni arc (Papua New Guinea, *McInnes et al.* 2001; *Franz et al.*, 2002).

58 Widespread metasomatic evidences have been documented in the Patagonian sub continental
59 lithospheric mantle by various authors. In many cases mantle xenoliths entrained in the back-arc
60 Patagonian lavas from various localities (between 40°S and 52°S) record regional, pervasive re-
61 crystallisation leaving only a few relics of the preceding mantle texture. Cryptic [trace element
62 enrichments of clinopyroxenes (cpx)] and modal (crystallization of hydrous phases such as amphibole ±
63 phlogopite) metasomatism occur in most of the studied suites. The point is to define the nature and to
64 understand the origin of the metasomatic melts that affect the mantle and that generate the observed
65 textural and geochemical features. Some authors attribute the metasomatic agent/s to silicate melts
66 similar to the host lavas as for Pali Aike (*Kempton et al.*, 1999; *Stern et al.* 1999). *Gorring and Kay*
67 (2000) provide information about a possible involvement of a carbonatite melt at Gobernador Gregores,
68 while a plume-related melt has been inferred by *Bjerg et al* (2005) for the Patagonian mantle. Slab
69 derived metasomatism has been proposed for one of the westernmost localities (i.e. the closest to the
70 trench) represented by Cerro del Fraile (*Kilian et al.*, 1998; *Kilian and Stern*, 2002; *Faccini et al.*,
71 2013), as well as for Cerro de los Chenques (*Rivalenti et al.*, 2007), Gobernador Gregores and Pali Aike
72 (*Stern et al.*, 1989; *Laurora et al.*, 2001).

73 In this study we present new major and trace element compositions of a suite of mantle xenoliths
74 sampled at Estancia Sol de Mayo (ESM), belonging to Meseta Lago Buenos Aires (MLBA, **Fig. 1B**),
75 which is one of the five back-arc Mesetas situated between 46° and 49°S. Our aim is to identify the
76 refertilization processes that could have affected this Central Patagonia locality and to constrain the
77 origin of the melts that have percolated into and interacted with the mantle.

78 **2. Geological setting**

79 In Patagonia the Andean volcanic arc is distinguished into a Southern Volcanic Zone (SVZ; *Thorpe et*
80 *al.*, 1982) and an Austral Volcanic Zone (AVZ; *Stern and Kilian*, 1996) separated by a volcanic gap
81 occurring between 46.30° and 49.00°S latitude. The geological history during the Cenozoic for both
82 SVZ and AVZ is related to the subduction of the Nazca (convergence rate of 10 cm*yr⁻¹) and Antarctic

83 (convergence rate of $2 \text{ cm} \cdot \text{yr}^{-1}$) plates beneath the South American plate. The two plates are separated
84 by the Chile ridge, and the present day position of the triple point between the Nazca, South American
85 and Antarctic plates (Chile Triple Junction, CTJ) occur at 46.30°S . (*Cande and Leslie, 1986; Forsythe et*
86 *al., 1986*).

87 A peculiar feature of Patagonia is the presence of several continental mafic volcanic plateaus ranging in
88 age from late Paleocene to Recent times (*Ramos and Kay, 1992*) (**Fig. 1A**). The sequence pre-plateaus –
89 main plateaus – post-plateaus is usually recognized, with the second stage being the most voluminous.
90 The Somoncuro igneous province, the largest post-Eocene mafic volcanic field of Northern Patagonia,
91 occurs between $\sim 40^\circ\text{S}$ and 46°S . It consists of a series of Oligocene to early Miocene volcanic fields
92 that cover more than $55,000 \text{ km}^2$ in the Meseta de Somoncuro and surrounding region (Meseta de Cari
93 Laufquen and Meseta de Canquel), overlying a late Precambrian to Paleozoic magmatic and
94 metamorphic basement itself covered by the extensive Jurassic silicic volcanic rocks of the Chon Aike
95 province (*Kay et al., 1989; Pankhurst and Rapela, 1995; Kay et al., 2007*), as well as Cretaceous to
96 Tertiary volcanic and sedimentary rocks (*Rapela and Kay, 1988; Rapela et al., 1988; Ardolino et al.,*
97 *1999*). Oligocene intraplate alkaline basalts and hawaiites are typical of the pre-plateau stage, followed
98 by a voluminous $\sim 27 \pm 2 \text{ Ma}$ hyperstene-normative basalt and basaltic andesite plateau sequence and by
99 intermediate to low volume post-plateau alkali olivine basalts and hawaiites ($\sim 23\text{-}17\text{Ma}$) (*Kay et al.,*
100 *2007*).

101 In Central Patagonia (between 46°S and 49°S) the middle Miocene to Recent northward migration of
102 the CTJ from approximately 50°S (*Cande and Leslie, 1986; Forsythe et al., 1986*) to 46.30°S has
103 generated unique geodynamic, structural and magmatic features (*Gorring et al., 1997*), namely the
104 modern volcanic arc gap between the SVZ and the AVZ, the eruption of arc adakitic magmas (*Kay et*
105 *al., 1993*) and finally the extensive late Miocene to Pleistocene magmatism that originated the Triple
106 Junction Province (TJP). It can be subdivided into a voluminous, late Miocene to early Pliocene main
107 plateau sequence, and a less voluminous, latest Miocene to Plio-Pleistocene post-plateau sequence
108 (*Gorring et al., 1997*). The main plateau sequence forms the smaller mesetas to the northeast (called
109 “northeastern region”) and the large and elevated plateaus of the de la Muerte (MM), Belgrano (MB),

110 Central (MC) and Lago Buenos Aires (MLBA) Mesetas (**Fig. 1B**). The post-plateau sequence comprises
111 small scoria cones, as well as lava flows and pyroclastic deposits capping the main plateau sequence.
112 OIB-like tholeiitic main plateau (~12-5 Ma) and alkaline post-plateau lavas (~7-2 Ma) are related to the
113 slab window tectonic evolution (*Ramos and Kay, 1992; Kay et al., 1993; Gorring et al., 1997*).
114 Finally between 49°S and 52°S (i.e. Southern Patagonia) there is the occurrence of the southernmost and
115 youngest (~3.8 Ma to Recent, *D'Orazio et al., 2000*) Cenozoic back-arc Patagonian lavas, represented
116 by the Pali Aike Volcanic Field (PAVF), being characterized by alkaline and olivine basalts and
117 basanites. It covers an area of about 4,500 km² north of the Magallanes fault system and is situated 200
118 km east of the Andean Cordillera. More than 80% of the totality of the volcanic products consists of an
119 extensive succession of plateau-like basaltic lava flows, while the remaining 20% consists of more than
120 450 monogenetic structures represented by maars, tuff-rings, scoria and spatter cones, and associated
121 lava flows (*D'Orazio et al., 2000*). *D'Orazio et al. (2000)* observed two main elongation trends of the
122 cones, one with an ENE direction and another with a NW direction, the first being linked to the still
123 active Magallanes Strait Rift System described by *Diraison et al. (1997)* while the second is probably
124 connected with the Mesozoic Patagonian Austral Rift (*Corbella et al., 1996*).

125

126 **3. Analytical methods**

127 This study is based on the major and trace element characterization of the mineral phases of Patagonian
128 mantle xenoliths carried out with an Electron Microprobe (EMP) and a Laser Ablation Inductively
129 Coupled Plasma Mass Spectrometer (LA-ICP-MS). Major and trace element compositions of the
130 entraining lavas were performed with X-Ray Fluorescence (XRF). All the analysis were performed at
131 the UMR 5563 (LMTG, Observatoire Midi-Pyrenees) of the University Paul Sabatier (Toulouse III),
132 except the bulk rock composition of the lava, performed at the University of Ferrara.
133 Major element compositions of minerals were determined with the CAMECA SX50 electron
134 microprobe and a standard program: beam current of 20 nA and an acceleration voltage of 15 kV, 10 –
135 30 s of peak counting, 10 s of background counting, and natural and synthetic minerals as standards.

136 Nominal concentrations were subsequently corrected using the PAP data reduction method (*Pouchou*
137 *and Pichoir 1984*). The theoretical lower limits of detection are about 100 ppm (0.01%).

138 Concentrations of REE and trace elements in cpx and orthopyroxene (opx) were determined in situ
139 using the Agilent 7500 ICP – MS instrument coupled either with CETAC laser ablation module that
140 uses either a 266 nm frequency-quadrupled Nd-YAG laser or a commercial femtosecond Ti : Sa laser
141 system (Amplitude Technologies Pulsar 10) based on the Chirped-pulse amplification (CPA) technique.
142 Pulses were amplified in this set-up by a regenerative and a multipass amplifier up to 12 mJ. This
143 system provides laser pulses at 800 nm with variable pulse energy and pulse duration as short as 50 fs.
144 Its contrast on 10 ps is on the order of 10^{-7} . Its repetition rate can be varied between 1 Hz and 10 Hz.
145 The shot-to-shot stability (RMS) is 2 %. The linearly polarized laser beam is injected in a BX51
146 microscope (Olympus). The beam is reflected by a 45° dielectric mirror and focused down to the sample
147 placed in an ablation cell mounted on an XY stage, using a 0.9 Cassegrain objective. The NIST 610 and
148 NIST 612 glass standards were used to calibrate relative element sensitivities of cpx and opx,
149 respectively (provided as supplementary data). Precision and accuracy (<5% and <20% respectively)
150 were assessed from repeated analyses of NIST 612 and NIST 610 as unknowns. Each analysis was
151 normalized using CaO and SiO₂ values, first determined by electron microprobe, for cpx and opx
152 respectively. A beam diameter of 50 – 100 μm and a scanning rate of 20 μm/s were used. The
153 theoretical limits of detection range from 10 – 20 ppb for REE, Ba, Th, U, Zr to 2 ppm for Ti..

154 Whole-rock major elements and some trace elements (Zn, Cu, Sc, Ga, Ni, Co, Cr, V, Rb, Ba, Th, Nb,
155 Sr, Zr, and Y) were obtained by X-ray fluorescence (XRF) on pressed-powder pellets, using an ARL
156 Advant-XP automated X-ray spectrometer. Calibration was performed using international reference
157 samples (some of which were also run as unknowns in order to determine accuracy and detection
158 limits), and the matrix correction method proposed by *Lachance and Traill (1966)* was applied. Mean
159 accuracies were generally better than 2% for major oxides, and better than 5% for trace element
160 determinations, while the detection limits for trace elements were: Zn, Ba, Cu, Sc = 5 ppm; Ni, Co, Cr,
161 V, Rb, Y, Th, Nb = 1 ppm; and Sr, Zr, Ga = 2 ppm. Volatile contents were determined to be lost on
162 ignition at 1000 °C.

163

164 **4. Chemical composition of the host lavas**

165 Six alkaline post-plateau host rocks have been analysed in this work. The samples do not represent an
166 extensive ad hoc sampling of MLBA. Previous works on the Plio-Pleistocene post-plateau volcanism in
167 this area (*Gorring et al., 2003*) link its origin to the opening of an astenospheric window during the
168 subduction of the Chile Ridge beneath the Andean margin ~6 Ma ago.

169 The lavas analysed are quite fresh, characterized by a porphyritic texture with 2 to 5% phenocrysts.
170 They are overwhelmingly dominated by euhedral olivine (ol), sometimes occurring in glomerophyric
171 assemblages (in some cases surrounded by a marked rim of reaction) that, since they carry mantle
172 xenoliths, should be of mantle origin. The groundmass is microcrystalline indicating a rapid magma
173 cooling, with abundant acicular plagioclase, associated with cpx, ol and Fe-Ti oxides.

174 The geochemical composition of MLBA lavas analyzed in this study is given in **Table 1**. MLBA post-
175 plateau lavas are sodic alkaline (~50 wt.% SiO₂; ~8 wt.% Na₂O+K₂O, with Na₂O/ K₂O >2), plotting in
176 the basaltic trachy-andesite field on a Total Alkali–Silica classification diagram (**Fig. 2**). They have low
177 MgO content (~4 wt. %) as well as Ni and Cr, the first being ~45 ppm and the second varying from 164
178 to 210 ppm.

179 Chondrite-normalized trace element concentrations of the samples are shown in **Fig. 3**. The patterns of
180 the lavas in this work resemble those of the OIB, as well as those of the main and post-plateau from the
181 TJ province, the latter having slightly higher incompatible trace element concentrations with respect to
182 those of the main plateau. The OIB signature of the samples is also highlighted by the low Ba/Nb ratios
183 that are less than 10 (typical of the composition within plate lavas).

184

185 **5. Petrography of the samples**

186 Fifteen xenoliths occurring in the post-plateau lavas of the MLBA have been studied. Most of them are
187 very small in size (a few centimeters across) and are rounded in shape. Their modal composition has
188 been calculated by counting over 1,000 points per thin section.

189 The xenoliths are mainly represented by anhydrous spinel-bearing harzburgites (7) and dunites (5), with
190 minor lherzolites (2) and one wehrlite (**Fig. 4**). They are characterized by a coarse grained protogranular
191 texture (*Mercier and Nicolas, 1975*) and they are devoid of metasomatic features, such as spongy rims,
192 reaction rims around spinel (sp) and/or opx, glassy patches, as well as any hydrous minerals.
193 Common textural features are the presence of two kinds of cpx and sp, as well as three types of opx. The
194 former generally occur as protogranular in the peridotitic matrix (cpx1, **Fig. 5A**) or growing around the
195 sp (cpx2, **Fig. 5B**); in this case the sp is identified as “sp_{cpx}” (**Fig. 5B**), whereas when separated from the
196 cpx it is called “sp”. Opx is present as i) large protogranular crystals with exsolution lamellae of cpx
197 (opx1, **Fig. 5A**), ii) small clean and undeformed grains without exsolution lamellae (opx2, **Fig. 5B**) and
198 iii) as smaller grains arranged in vein (opx3, **Fig. 5C**).

199 **6. Geochemistry of the mineral phases**

200 **6.1. Major elements**

201 **6.1.1. Olivine**

202 Ol of lherzolites have a Fo content ranging from 90.5 to 91.3 (**Table 2**). Ol in the two lherzolite samples
203 are characterized by a wide range of SiO₂ content, with MGP2b2 (39.5-42.1 wt. %) showing a higher
204 variation than MGP2b (40.2-41.4 wt. %), and by a similar range of NiO (0.27-0.50 wt. %). Ol in
205 harzburgites have Fo varying from 84.2 up to 92.1. Apart from sample MGP4b, which presents a Fo
206 content (91.7-92.1) higher than those of lherzolitic ol and a narrow range of variation of SiO₂ (40.9-41.7
207 wt. %) and NiO (0.33-0.45 wt. %), the remaining harzburgites have similar or lower Fo content (89.2-
208 90.8). Ol of harzburgites MGP1b and MGP1g have particularly low Fo (84.2-88.6) as well as SiO₂ and
209 NiO contents, the former varying from 39.2 to 40.4 wt. % and the latter from 0.17 to 0.46 wt. %
210 respectively. Ol of dunites (MGP1h and MGP2a) are characterized by Fo (89.4-91.4) contents akin to
211 those of lherzolites with sample MGP1h (90.1-91.4) showing a range slightly higher than the one of
212 MGP2a (89.4-90.1). Also the SiO₂ and NiO content variations are similar to those of lherzolites, varying
213 from 39.6 to 41.2 wt. % and from 0.21 to 0.46 wt. %, respectively. Finally, ol of the wehrlite shows the

214 lowest Fo values, ranging from 81.3 to 82.1. They also show low SiO₂ and NiO values, ranging,
215 respectively, from 38.9 to 39.7 wt. % and 0.14 to 0.20 wt. %.

216 6.1.2. Clinopyroxene

217 Cpx are classified according to their textural features i.e. those occurring as protogranular (cpx1) and
218 those linked to sp (cpx2) (**Fig. 5A and B**).

219 Cpx1 and cpx2 from lherzolites have similar mg#, the former ranging from 91.0 to 92.0 and the latter
220 between 90.6 and 91.9. Cpx2 shows slightly higher Al₂O₃ and Cr₂O₃ (**Table 3 and Fig. 6A**) contents
221 (4.01-4.31 wt. % and 1.19-1.45 wt. %, respectively) with respect to cpx1 (3.21-4.12 wt. %, with one
222 reaching 4.52 wt. %, and 0.56-1.02 wt. %, respectively). Both types of cpx have similar CaO (20.9-21.6
223 wt. %), Na₂O (0.76-1.02 wt. %) and TiO₂ (0.12-0.39 wt. % with a couple of cpx1 having 0.05 and 0.08
224 wt. %) contents, whereas SiO₂ varies widely, with cpx2 marked by lower values (50.0-53.1 wt. % for
225 cpx2 and 51.1-53.7 wt. % for cpx1).

226 The mg# of cpx1 from harzburgites varies from 89.9 up to 92.9; they are also characterized by higher
227 SiO₂ (51.8-54.1 wt. %) contents than those of the lherzolites, lower Al₂O₃ (1.75-3.56 wt. %) as well as
228 TiO₂ (0.05-0.2 wt. %) contents, but similar Cr₂O₃ (0.57-1.45 wt. %), Na₂O (0.71-1.0 wt. %), and CaO
229 (20.8-21.7 wt. %) contents. All cpx2 (except those belonging to harzburgite MGP1b [mg# 87.9-88.9])
230 show a range of mg# (91.5-92.3), Al₂O₃ (3.46-4.45 wt. %), Cr₂O₃ (1.26-1.59 wt. %), CaO (21.0-21.3),
231 and TiO₂ (0.12-0.31 wt. %) akin to those of cpx2 from lherzolite, showing a narrower range of SiO₂
232 (52.3-53.1 wt. %) and a higher content of Na₂O (0.95-1.18 wt. %). Cpx2 from harzburgite MGP1b do
233 not show any compositional differences with the cpx1 from the same sample.

234 Cpx1 from dunites have a very narrow mg# range, from 89.5 to 91.5 similar to that in harzburgite
235 MGP3b. They also display the same Al₂O₃ (1.75-3.12 wt. %) but with a slightly lower Cr₂O₃ (from 0.73
236 to 1.20 wt. %) content. At comparable mg# with harzburgites, they are characterized by higher TiO₂
237 (0.07-0.30 wt. %) and Na₂O (0.74-1.13 wt. %) contents (but similar to those of lherzolites) and lower
238 SiO₂ (52.1-53.4 wt. %) and CaO (20.5-21.8 wt. %) contents. With respect to cpx1, cpx2 have mg#
239 (89.6-90.9) similar to that of the cpx1, but have higher Al₂O₃ (2.65-4.03 wt. %) and Na₂O (1.00-1.27 wt.
240 %), slightly higher Cr₂O₃ (0.91-1.36) and TiO₂ (0.24-0.51 wt. %) and slightly lower SiO₂ and CaO

241 contents (varying from 51.6 to 52.8 and from 19.9 to 21.1, respectively). All these features collocate the
242 cpx2 of dunites in the same field as those previously described for lherzolites and harzburgites (with the
243 exception of cpx2 from harzburgite MGP1b).

244 Cpx of the wehrlite show the lowest mg# ranging from 81.5 to 84.6. They are characterized by the
245 highest Al₂O₃ and TiO₂ contents (5.13 - 7.3 wt. % and 0.65 - 1.17 wt. %, respectively) and the lowest
246 Cr₂O₃ (0.37-0.86 wt. %), CaO (15.75-21.4 wt. %), Na₂O (0.63-1.00 wt. %) and SiO₂ (49.4-51.6 wt. %)
247 contents.

248 6.1.3. Orthopyroxene

249 Opx have been divided in three groups on the basis of their petrographic and textural features (**Fig. 5A,**
250 **B** and **C**). No differences between opx1 and opx2 have been found in terms of major element
251 composition, whereas opx arranged in vein (opx3, dunite MGP2a) is always quite well discriminated
252 chemically, too.

253 Opx from lherzolites have mg# ranging from 90.1 to 91.5 with an Al₂O₃ content ranging from 2.62 to
254 2.97 wt. % (**Table 4, Fig. 6B**). They also show high and quite variable Cr₂O₃ (0.48-0.70 wt. %), SiO₂
255 (55.3-56.4 wt. %) and CaO (0.84-0.99 wt. %) contents. Opx of harzburgites are characterized by a wider
256 mg# range varying from 84.2 to 93.0. Two samples (harzburgites MGP3b and MGP1c) have mg# akin
257 to those of lherzolites (90.1-91.6), with lower Al₂O₃ (1.37-2.14 wt. %), slightly lower Cr₂O₃ (0.26 to
258 0.63 wt. %), similar CaO [0.82 and 0.97 wt. % (one grain belonging to harzburgite MGP1d with 0.72
259 wt. %)] and higher SiO₂ contents (55.8-57.1 wt. %). Opx from the other two harzburgites (MGP1b and
260 MGP1g) have Al₂O₃ (2.30-3.32 wt. %) contents comparable (or slightly higher) than those of lherzolites
261 but they have lower mg#, ranging from 84.2 to 88.7. They are also marked by similar CaO contents
262 (0.79-0.98 wt. %) but lower SiO₂ (54.1-55.9 wt. %) and Cr₂O₃ (0.26-0.60 wt. %) contents. Finally opx
263 from the harzburgites MGP4b are characterized by a very high mg# (91.6-93.0), and by Al₂O₃ (2.38-
264 2.94 wt. %), SiO₂ (55.8-57.2 wt. %) and CaO (0.85-1.00 wt. %) similar to those of lherzolites. Cr₂O₃
265 content is slightly higher, ranging from 0.44 to 0.73 wt. %.

266 Opx3 (those arranged in vein in the dunite MGP2a) are always distinguished from the other two kinds
267 of opx. Mg# is analogous to those of opx from lherzolites and harzburgites MGP1c and MGP3b,

268 comprised between 89.2 and 90.6, but, at comparable mg#, they have (higher Al₂O₃ (3.02-3.52 wt. %),
269 and lower CaO (0.76-0.89 wt. %), SiO₂ (54.0-55.5 wt. %) and Cr₂O₃ (0.20-0.40 wt. %) contents. Opx1
270 from the other dunitic sample (MGP1h) have similar Cr₂O₃ (comprised in the range 0.25-0.42 wt. %,
271 except one analyses, up to 0.56 wt. %) and CaO (0.80-0.96 wt. %) contents to those of opx3 of dunite
272 MGP2a, but this latter one shows higher SiO₂ (57.3-57.9 wt. %) and lower Al₂O₃ (1.18-1.59 wt. %)
273 contents.

274 6.1.4. Spinel

275 On the basis of their petrographic and textural features, sp have been classified in two groups: those
276 occurring with cpx (sp_{cpx}) and those not related to pyroxene (sp). The first group is composed solely of
277 sp_{cpx} that have higher Al₂O₃ content with respect to Cr₂O₃ (mg# ranging from 61.2 to 77.2 and cr#
278 between 37.1 and 51.0). The other sp group can be characterized by higher cr# (49.2-60.6) and mg#
279 comprised between 62.0 and 67.4. A few grains of both sp and sp_{cpx} plot outside these two groups, at
280 very low cr# (17.5-26.5) and mg# (55.0-64.9).

281 Taking into account the lithotype, sp from lherzolites have mg# and cr# ranging from 72.3 to 73.1, and
282 from 37.1 to 38.2 respectively, with only one grain of sample MGP2b2 falling outside these ranges
283 (mg#=77.2 and cr#=38.5) (**Table 5**). Those belonging to harzburgites have a quite different
284 geochemical composition, with one sample (MGP4b) having the highest mg# (74.8-76.9) and cr# (38.3-
285 40.5) similar to that of lherzolites; three harzburgites (MGP1b, MGP1c and MGP3b) are characterized
286 by higher cr# with respect to harzburgite MGP4b and two lherzolites (42.6-60.0) and lower mg# (61.2-
287 67.5), and two sp of harzburgite MGP1g with a very low mg# and cr#. Sp of dunites fall in the field
288 defined by the three harzburgites MGP1b, MGP1c and MGP3b, with mg# ranging from 62.0 to 66.4 and
289 cr# from 46.6 to 57.1. Three sp_{cpx} from dunite MGP1h plot outside this group, having slightly higher
290 mg# (69.6-70.4) and lower cr# (43.5-43.9). Finally sp of the wehrlite are the most aluminiferous (cr#
291 17.5-20.8) with mg# ranging from 62.6 to 64.9.

292 6.2. Trace elements

293 6.2.1. Clinopyroxene

294 Chondrite-normalized (*Sun and McDonough, 1989*) trace elements and rare earth element (REE)
295 compositions for each sample are reported in **Fig. 7** and **Table 6**. No correlation between composition
296 and textural position and/or the lithotype have been observed for cpx1 and cpx2. They are practically
297 indistinguishable solely on the basis of trace element composition.

298 Cpx from lherzolites are characterized by a remarkable Th positive anomaly, a strong Nb and less
299 pronounced Zr and Ti negative anomalies. Most of them are light REE (LREE) enriched with $(La/Yb)_N$
300 ranging from 2.09 to 5.57. A few grains are characterized by a lower $(La/Yb)_N$ ratio comprised between
301 1.55 and 1.96, related to an increase of the heavy REE (HREE) content. Cpx from harzburgites show
302 anomalies similar to those highlighted for the lherzolites. With respect to these latter they are
303 characterized by a more pronounced negative Ti anomaly and by a variable, but always negative, Zr
304 anomaly. Harzburgite MGP1c (and a few grains of MGP4b) show a slightly negative Zr anomaly,
305 whereas cpx from harzburgites MGP1b, MGP3b and MGP4b are marked by a strong negative Zr
306 anomaly. The REE patterns resemble those of the lherzolites, with an enrichment in LREE, most of the
307 cpx having a $(La/Yb)_N$ comprised between 2.09 and 7.28. As for lherzolites, some cpx in harzburgites
308 are characterized by HREE enrichment, leading to a decrease of the $(La/Yb)_N$ ratio (0.87-1.68).

309 Only one cpx crystal was found and analysed in dunite MGP2a. It shows a more fractionated
310 incompatible trace element pattern, always characterized by the prominent positive Th and negative Nb
311 anomalies. It also has marked Zr and Ti negative anomalies, and a steep REE pattern, with a $(La/Yb)_N$
312 equal to 4.11.

313 Cpx of the wehrlite are also characterized by negative Nb and positive Th anomalies, even if the latter is
314 less marked than that of the other cpx. Indeed, in lherzolites Th content varies from 0.24 up to 14.7
315 ppm, in the harzburgites from 0.71 to 14.2 ppm, in the dunite it is equal to 2 ppm, whereas in wehrlite it
316 ranges from 0.12 and 0.41 ppm. The two negative Zr and Ti anomalies are present in the wehrlitic
317 clinopyroxenes, the former anomaly being generally marked and the second varying from slight to
318 strong. REE patterns highlight two different compositions, one with a convex upward pattern and a
319 $(La/Yb)_N$ ranging from 0.79 and 1.04 and the other with LREE enrichment and a $(La/Yb)_N$ varying from
320 2.87 up to 5.26. These two compositions, however, are not related to different textural position.

321 6.2.2. Orthopyroxene

322 Chondrite-normalized (*Sun and McDonough, 1989*) trace elements profiles and REE of opx are reported
 323 in **Fig 8** and **Table 7**. Lherzolites are characterized by a prominent positive Th and negative Sr, Zr and
 324 Ti anomalies. They also show depleted LREE with a negative Ce anomaly and HREE at about 5X
 325 chondritic, with $(La/Yb)_N$ ratios varying between 0.06 and 0.30.

326 Opx from harzburgites MGP1b and MGP1c display the same anomalies observed for the lherzolites, i.e.
 327 positive Th and negative Sr, Zr and Ti anomalies as well as flat HREE with a drastic LREE depletion
 328 resulting in a $(La/Yb)_N$ ratio ranging from 0.14 to 0.74. The other three harzburgites (MGP1g, MGP3b
 329 and MGP4b) are characterized by the same strong positive Th and negative Sr anomalies but slightly
 330 negative Zr and Ti anomalies. The REE patterns show flat HREE and depleted LREE, with $(La/Yb)_N$
 331 ranging from 0.16 and 0.74.

332 Opx analysed in the two dunites MGP1h and MGP2a always show marked positive Th and negative Sr
 333 anomalies, but the former has slightly negative Zr and Ti anomalies, the latter (opx3 in vein) displays a
 334 positive Zr and a prominent negative Ti anomaly. The REE patterns are similar to those described
 335 above, with flat HREE and a depletion in LREE also resulting in this case in the low $(La/Yb)_N$ ratio,
 336 between 0.21 and 0.36. Considering the different REE patterns and the negative Ti anomaly, Ti^*
 337 $[Ti_N / ((Eu_N + Gd_N) / 2)]$ is plotted versus $(Ce/Yb)_N$, highlighting the presence of two groups of opx, one at
 338 low Ti^* (i.e. prominent Ti negative anomaly) and $(Ce/Yb)_N$ and one at higher values of both $(Ce/Yb)_N$
 339 and Ti^* (**Fig. 9A**). **Fig. 9B** better constrains this subdivision by taking into account the Zr negative
 340 anomaly (Zr^* , $[Zr_N / ((Sm_N + Nd_N) / 2)]$) which varies from slight to prominent, but also positive in one
 341 case. Moreover it allows distinguishing a third group constituted by the opx in the vein of sample
 342 MGP2a that clearly plots outside Group II due to its positive Zr anomaly.

343 7. P-T conditions and fO_2 .

344 Temperature and Pressure conditions were estimated using the two-pyroxene thermometer of *Brey and*
 345 *Kölder (1990)* and the *Kölder and Brey (1990)* barometer, this latter one based on the Ca exchange
 346 between ol and cpx. Temperature has been estimated on lamellae-free opx. Some care is needed

347 regarding the pressure estimates because the Ca content of the ol has only been measured by electron
348 microprobe. Beside some unreasonably high (>30 Kb) and low (<5 Kb) pressure values, most of the
349 samples fall in the spinel stability field (P ranging from 12 to 20 Kb) within a narrow range of
350 temperature (1003 °C – 1040 °C).

351 Temperature estimates for ESM compare favourably to the upper ranges of estimates from other Central
352 Patagonia localities. *Bjerg et al. (2005)* calculated at Gobernador Gregores equilibration temperatures
353 for spinel peridotites ranging from 830 °C to 1090 °C; *Dantas (2007)* reports temperatures from 850 °C
354 to 1100 °C for the same locality. Temperatures of spinel lherzolites from Tres Lagos range between 728
355 °C and 1040 °C (*Ntaflos et al., 2006*), while *Faccini et al. (2013)* calculates temperatures between 872
356 °C and 1006 °C at Cerro Fraile.

357 $\Delta\log f_{O_2}$ (QFM) have been calculated using the equilibrium of the ol-sp-opx assemblage according to
358 *Ballhaus et al. (1991)* (**Fig. 10**). Pressure used for the oxygen fugacity is 15 Kbar, while temperatures,
359 varying between 912 °C and 980 °C, are in good agreement with those calculated with the *Brey and*
360 *Kölder (1990)* thermometer (**Table 8**). All samples have positive $\Delta\log f_{O_2}$ indicating oxidized
361 conditions. The two lherzolites have the lowest values (+0.02 and +0.62) while the harzburgites have
362 values higher than +1, with the most oxidized samples reaching +1.47. The mean value of the f_{O_2} is
363 +0.86, slightly higher than that calculated for supra-subduction mantle xenoliths (+0.51, *Foley, 2011*).
364 The oxygen fugacity calculated at ESM is quite different from that calculated at Cerro del Fraile. For
365 this locality, an average $\Delta\log f_{O_2}$ of -0.41 and -0.34 were calculated by *Faccini et al. (2013)* and *Wang*
366 *et al. (2007)* respectively.

367

368 **8. Discussion**

369 **8.1. Depletion processes**

370 Partial melting of ESM mantle xenoliths has been estimated using trace (*Johnson et al., 1990*) and
371 major element (*Bonadiman et al., 2011*) modelling, the former based on the HREE content in cpx and
372 the second on the Al₂O₃ content in cpx and opx.

373 It is well known that HREE in peridotites are very sensitive to the partial melting degree (F), i.e. an
374 increase of F corresponds to a decrease of HREE in the residue. Contrary to LREE, HREE values are
375 also less affected by successive enrichment due to metasomatism. In **Fig. 11A** and **B** the melting curves
376 for REE (with Zr, Ti and Y) are reported according to the equations of *Johnson et al. (1990)*, starting
377 from a Primitive Mantle (PM) composition from *Bonadiman et al. (2005)* and assuming a modal
378 composition of 55% ol, 22% opx, 20% cpx and 3% sp, at 1.5 GPa. Cpx HREE contents in lherzolites
379 suggest an F between $\sim 5\%$ and $\sim 13\%$ (**Fig.11A**); cpx from harzburgite MGP3b and few cpx from
380 MGP4b fall in the range of F comprised between $\sim 5\%$ and $\sim 15\%$ (**Fig. 11B**). The cpx of the other
381 harzburgites (MGP1b and MGP1c) have HREE equal or higher than PM and for this reason they do not
382 lead to any result.

383 The melting history of the ESM mantle is also provided by the Al_2O_3 variation for cpx and opx,
384 according to the melting trends of *Bonadiman et al. (2011)* (**Fig. 12**). Al_2O_3 behavior is strictly related
385 to partial melting processes in basaltic systems, i.e. the increasing of melting results in a decreasing of
386 the Al_2O_3 in opx, cpx and sp, and for this reason it can be chosen as a robust geochemical parameter
387 when dealing with partial melting modelling (*Ionov and Hofmann, 2007; Bonadiman et al., 2011*). In
388 the Al_2O_3 vs. MgO diagram (**Fig. 12A** and **B**) an F comprised between $\sim 15\%$ and $\sim 30\%$ for the opx
389 (**Fig. 12A**) and between $\sim 16\%$ and $\sim 23\%$ for cpx (**Fig. 12B**) is indicated, with opx being more
390 scattered than cpx.

391 Opx from the two lherzolites record the same degree of melting ($F \sim 20\%$) while the cpx spans from \sim
392 16% to 20%. Opx from harzburgites are distributed quite well along the partial melting line in both
393 cases, even if those from MGP1c tend to diverge towards lower and higher values of MgO and Al_2O_3
394 respectively and those of MGP4b towards higher values of both oxides. In this case the estimated F
395 ranges from $\sim 20\%$ to $\sim 30\%$. On the contrary, cpx of harzburgites are well aligned on the curve,
396 recording an F between $\sim 16\%$ and $\sim 23\%$. Cpx from dunites are slightly scattered towards lower Al_2O_3
397 values aligning along a partial melting curve parallel to that of *Bonadiman et al. (2011)* with F similar
398 to that of the cpx from harzburgites ranging from $\sim 16\%$ to $\sim 23\%$. Opx from MGP2a recording an F

399 similar to that of the lherzolites (~ 16%) should not be considered because it is arranged in vein and it
400 clearly postdates the partial melting event(s).

401 To summarize, HREE modelling of cpx from lherzolites record an F varying between ~ 5% and ~ 13%.
402 The highest values (F ranging from ~ 16% and ~ 23%.) obtained through the Al_2O_3 model are probably
403 too high to be reliable, also taking into account the fact that the cpx modal content of this lithotype
404 varies from 9.5% (MGP2b) to 13.3% (MGP2b2). On the other hand, HREE content of cpx from
405 harzburgites MGP3b and MGP4b record an F comprised between ~ 5% and ~ 15%. In this case the
406 highest values (F comprised between ~ 20% and ~ 30%.) obtained by the major element modelling are
407 compatible with the modal composition of 73% ol, 21% opx, 4% cpx and 2% sp.

408 8.2. Enrichment processes and nature of the incoming melt(s)

409 Cpx and opx of ESM are characterized by a decrease in Al_2O_3 content related to an increase in mg#
410 (Fig. 6A and B). Furthermore a correlation between the increase of the mg# and the nature of the
411 lithotype, i.e. the modal content of cpx left after the partial melting (a gradual shift from lherzolites to
412 harzburgites to dunites) is not observed. An evolution of the mantle beneath ESM linked only to a
413 partial melting event is also ruled out by the negative correlation between major (in terms of Al_2O_3
414 contents) and trace element compositions (namely LREE and Sr, Fig. 13A and B). A residue after
415 partial melting in fact would have minor Al_2O_3 content coupled with a decrease in LREE (i.e. a decrease
416 of the La/Yb ratio) and Sr (i.e. a positive correlation), contrary to what was observed at ESM. These
417 geochemical features, together with the petrographic evidence of two texturally different cpx and the
418 presence of a vein of recrystallized opx, highlight the occurrence of possible metasomatic and/or
419 refertilization events affecting the upper mantle beneath ESM.

420 The main problem when dealing with metasomatism/refertilization events is the identification of the
421 liquid percolating through the mantle thereby modifying its geochemical features. In order to constrain
422 the nature of the melt, we compared the incompatible trace elements and REE patterns of the ESM cpx
423 with those of pyroxenites and wehrlites from northern and central Patagonia (Dantas, 2007) (Fig. 14A-
424 F). Four main processes are accounted for the generation of pyroxenites. Various authors consider them

425 as (1) formed by oceanic crust recycling (*Polvé & Allègre, 1980; Allègre & Turcotte, 1986; Kornprobst*
426 *et al., 1990; Kerr et al., 1995*), or (2) as cumulus at the base of the magmatic chamber (*DeBari &*
427 *Coleman, 1989; Schiano et al., 2000*), as well as (3) segregated at high pressure from mafic silicate
428 liquids (*Downes, 2005*) or (4) as products of the interaction between the peridotite with melts at mantle
429 depth, suggesting the presence of refertilization/metasomatic event(s) (*Kelemen et al., 1992; Smith &*
430 *Riter, 1997; Wilkinson & Stolz, 1997; Garrido & Bodinier, 1999; Smith et al., 1999; Zanetti et al., 1999;*
431 *McInnes et al., 2001; Wang et al., 2001*). In this case, the samples are porphyroclastic to equigranular
432 opx-rich websterites from Cerro Rio Chubut and Cerro Aznare (**Fig. 1**, Northern Patagonia), as well as
433 porphyroclastic to equigranular olivine and spinel websterites from Cerro de Los Chenques (**Fig. 1**,
434 Northern Patagonia), and porphyroclastic spinel clinopyroxenites from Cerro Clark (**Fig. 1**, Central
435 Patagonia).

436 The first group of pyroxenites from Northern Patagonia is characterized by prominent Nb and slightly
437 negative Ti anomalies, depleted LREE and enriched-to-flat HREE (**Trend 1, Fig.14A and B**). The same
438 feature can also be observed for the cpx of the pyroxenites from central Patagonia that fall in the same
439 group, except for a slightly negative Zr anomaly and for less enriched REE patterns. Wehrlites from
440 central Patagonia show the highest trace element concentrations, with prominent negative Nb and Ti
441 anomalies, a negative to positive Zr anomaly, enriched LREE and fractionated HREE (**Trend 2,**
442 **Fig.14C and D**). Finally, a second group of cpx of pyroxenites from northern Patagonia is characterized
443 by a wider range of trace element concentrations, with negative Nb, Zr and Ti anomalies. LREE
444 contents vary from depleted to slightly enriched and fractionated to flat HREE (**Trend 3, Fig.14E and**
445 **F**). The grey field in **Fig. 14**, corresponding to the ESM cpx, resembles the pattern of cpx of **Trend 3**.

446 To constrain the origin of **Trend 1**, the REE patterns of cpx belonging to the pyroxenites of Northern
447 and Central Patagonia have been compared to those of cpx phenocrysts from tholeiitic lavas from
448 Ethiopian Rift (*Beccaluva et al., 2009*) (**Fig. 15**). To the best of our knowledge no trace element
449 analyses are available on cpx phenocrysts from Patagonian lavas. The overlap between cpx in
450 equilibrium and Ethiopian tholeiitic lavas is quite remarkable. This fact lends support to the percolation
451 of sub-alkaline SiO₂ – saturated melt beneath Northern and Central Patagonia. In the main plateau –

452 post-plateau (and in some cases also pre-plateau) eruption sequence typical of the Somoncuro and Triple
453 Junction mesetas, tholeiitic products are predominant especially in the voluminous main plateau
454 volcanic stage. It is thus likely that the northern and central Patagonian mantle suffered tholeiitic
455 refertilization processes, characterized by a much higher melt/rock ratio with respect to a metasomatic
456 event.

457 The cpx REE patterns of wehrlites belonging to **Trend 2** appear enriched in LREE with a La/Lu ranging
458 from 6.89 to 30.5 and a high content of Nb, Sr and Zr. These resemble typical patterns for mineral
459 crystallized from alkaline melts. According to Dantas (2007) these patterns are, in fact attributed to
460 CaO-rich SiO₂-undersaturated melt.

461 Incompatible trace elements and REE patterns of ESM cpx resemble those of pyroxenites from Northern
462 Patagonia generating **Trend 3** (**Fig. 14E and F**). The pattern of the cpx in equilibrium with the
463 transitional/alkaline lavas from the post-plateau stage of these provinces were calculated using the
464 appropriate partition coefficient (data from GERM, <http://earthref.org/GERM/>, provided as a
465 supplementary table) and considering the most and the least enriched lavas belonging to the Triple
466 Junction Province. As can be seen in **Fig. 16** cpx calculated (provided as supplementary data) in
467 equilibrium with the transitional/alkaline lavas from TJ province have patterns that are quite comparable
468 with those of the cpx from Northern Patagonia pyroxenites (**Trend 3**) and from ESM. This is also
469 supported by a favourable comparison between La/Yb ratios of cpx from ESM (comprised between 0.87
470 and 7.28), those of cpx from pyroxenites (between 0.26 and 6.95) and those of the calculated cpx
471 (between 0.57 and 6.44).

472 This supports the idea that a transitional alkaline/subalkaline melt refertilization event has affected the
473 mantle beneath ESM, as also suggested by the textural features represented by cpx₂ and opx in vein. A
474 few samples show an incongruent Al₂O₃ behavior between opx and cpx (**Fig. 6**) that can be explained
475 by an incomplete refertilization process. In fact cpx₂ and opx₃ tend to be enriched in Al₂O₃ with respect
476 to the cpx₁ and opx₁. This, together with the presence of small vein of opx, would point toward a
477 process occurring in recent time, just prior to xenolith entrainment.

478 It is to be pointed out that, while migrating through the mantle, transitional-type basalts can easily move
479 above and below the saturation threshold. Reaction and dissolution of opx would in fact increase the
480 SiO₂-saturation level, while its crystallization as completely new minerals or in substitution of ol would
481 decrease it (*Arai et al., 2006*) Within this framework we can place the slightly different, more alkaline,
482 pattern of the cpx in dunite MGP2a, as well as the presence of small opx in vein suggesting a SiO₂-rich
483 melt. Whether or not these affinities represent truly distinct families of melts or simple, small volume
484 variations of a unique transitional melt will be the topic of a forthcoming paper taking into consideration
485 the metasomatic/refertilization petrological modifications of the entire Patagonia (*Melchiorre et al., in*
486 prep.)

487 9. Conclusions

488 Anhydrous spinel-bearing peridotites (mainly harzburgites and dunites, with minor lherzolites and one
489 wehrlite) sampled at ESM (Patagonia), without any evidence of spongy rims or glassy patches, show
490 two and three texturally different cpx and opx, respectively. They depict two different trends, one
491 characterized by high Al₂O₃ content at almost constant mg# and the second by a slight increase of the
492 Al₂O₃ content with a decreasing of mg#. The trace element concentrations do not evidence any
493 difference between cpx1 and cpx2, but discriminate three groups of opx, in agreement with the observed
494 petrographical features: one is represented by the opx3 (those arranged in vein) characterized by a
495 prominent positive Zr anomaly, while the other two always show prominent-to-slightly negative Ti and
496 Zr anomalies and LREE depleted patterns.

497 Equilibration temperature estimates of ESM peridotites range from 1003 °C and 1040 °C, comparing
498 favourably to the upper ranges of estimates from other Central Patagonia localities. Positive $\Delta\log f_{O_2}$
499 (QFM, average of +0.86) is compatible with that calculated for a supra-subductive mantle by *Foley*
500 (*2001*) (+0.51), revealing an oxidizing environment, on the contrary of what was proposed by *Wang et*
501 *al (2007)* and *Faccini et al. (2013)* at Cerro del Fraile (-0.34 and -0.41 respectively). Major and trace
502 element modelling of partial melting reveals an F ranging from ~ 5% and ~ 13% for the lherzolites and
503 between ~ 20% and ~ 30% for the harzburgites.

504 The correlation between incompatible trace elements $[(La/Yb)_N \text{ and } Sr_N]$ and Al_2O_3 of the cpx highlight
505 the presence of a refertilization event affecting the ESM upper mantle, evidenced by the enrichment of
506 the LREE and Sr correlated to a decrease in the Al_2O_3 content. The agent that can account for this
507 process has a transitional affinity and is analogous to the lavas occurring within the various post-plateau
508 stages of the mesetas belonging to the Triple Junction Province. This conclusion has been reached by
509 reconstructing the REE patterns of a cpx in equilibrium with lavas with the lowest and highest trace
510 element contents from this province that resemble those of the cpx from ESM.

517 **Acknowledgements**

518 This work is part of a co-tutorship Ph.D. held within the 2008 Vinci Project “Matter transfer in supra-
519 subductive mantle in complex converging systems” funded by the “Università Italo-Francese”. The
520 activities have been carried out between the University of Ferrara and the CNRS of Toulouse. We thank
521 the laboratory staff of UMR 5563 (LMTG, Observatoire Midi-Pyrenees) for their support during the
522 performance of EMP and LA-ICP-MS analysis. We also thank Renzo Tassinari (University of Ferrara)
523 for the XRF analysis and Grant George Buffett (Institute of Earth Science Jaume Almera – CSIC of
524 Barcelona) for English editing. We thank Theodoros Ntaflos, Csaba Szabo and Sonja Aulbach for their
525 revisions. Their constructive comments have largely improved the first version of the manuscript.

526

527 **10. References**

- 528 Allègre, C. J., Turcotte, D. L., 1986. *Implications of a two-component marble-cake mantle. Nature* 323,
529 123-127.
- 530 Aoki, K., 1987. *Japanese Island arc: xenoliths in alkali basalts, high-alumina basalts, and calc alkaline*
531 *andesites and dacites, in P. H. Nixon (Ed), Mantle Xenoliths. John Wiley & Sons, New York, 319–33.*
- 532 Arai, S., Hirai, H., ABE, N., 1998. *Petrological characteristics of the sub-arc mantle: An overview on*
533 *petrology of peridotite xenoliths from the Japan arc. Trends in Mineralogy (India) 2, 39–55.*
- 534 Arai, S., Ishimaru, S., Okrugin, V. M., 2003. *Metasomatized harzburgite xenoliths from Avacha volcano*
535 *as fragments of mantle wedge of the Kamchatka arc: an implication for the metasomatic agent. Island*
536 *Arc* 12, 233-246.
- 537 Arai, S., Shimizu, Y., Morishita, T., Ishida, Y., 2006. *A new type of orthopyroxenite xenolith from*
538 *Takashima, Southwest Japan: silica enrichment of the mantle by evolved alkali basalt. Contribution to*
539 *Mineralogy and Petrology* 152, 387-398.
- 540 Ardolino, A., Franchi, M., Remesal, M., Salini, F., 1999. *El volcanismo en la Patagonia extrandina, in:*
541 *R. Caminos (Ed), Geologia Argentina. Servicio Geologico Nacional Anales* 29, 579–612.
- 542 Bali, E., Zajacz, Z., Kovács, I., Szabó, CS., Halter, W., Vaselli, O., Török, K., Bodnar, R. J., 2008. *A*
543 *quartz-bearing orthopyroxene-rich websterite xenoliths from the Pannonian Basin, Western Hungary:*
544 *evidence for release of quartz-saturated melts from a subducted slab. Journal of Petrology* 49, 421-439.
- 545 Ballhaus, C., Berry, R. F., Green, D. H., 1991. *High pressure experiment calibration of the olivine-*
546 *orthopyroxene-spinel oxygen barometer: implications for the oxidation state of the mantle.*
547 *Contributions to Mineralogy and Petrology* 107, 27-40.
- 548 Beccaluva, L., Bianchini, G., Natali, C., Siena, F., 2009. *Continental flood basalts and mantle plume: a*
549 *case study of the Northern Ethiopian Plateau. Journal of Petrology* 50, 1377-1403.
- 550 Berkesi, M., Guzmics, T., Szabó, C., Dubessy, J., Bodnar, R. J., Hidas, K., Ratter, K., 2012. *The role of*
551 *CO₂-rich fluids in trace element transport and metasomatism in the lithospheric mantle beneath the*
552 *Central Pannonian Basin, Hungary, based on fluid inclusions in mantle xenoliths. Earth and Planetary*
553 *Science Letters* 331-332, 8-20.

- 554 Bianchini, G., Beccaluva, L., Nowell, G.M., Pearson, D.G., Siena, F., 2011. Mantle xenoliths from
555 Tallante (Betic Cordillera): insights into the multi-stage evolution of the south Iberian lithosphere.
556 *Lithos* 124, 308-318.
- 557 Bjerg, E. A., Ntaflou, T., Kurat, G., Dobosi, G., Labudía, C. H., 2005. The upper mantle beneath
558 Patagonia, Argentina, documented by xenoliths from alkali basalts. *Journal of South American Earth
559 Sciences* 18, 125-145.
- 560 Bonadiman, C., Beccaluva, L., Coltorti, M., Siena, F., 2005. Kimberlite-like metasomatism and 'garnet
561 signature' in spinel peridotite xenoliths from Sal, Cape Verde Archipelago: relics of a subcontinental
562 mantle domain within the Atlantic Oceanic lithosphere? *Journal of Petrology* 46, 2465-2493.
- 563 Bonadiman, C., Coltorti, M., Beccaluva, L., Griffin, W. L., O'Reilly, S.Y., Siena, F., 2011.
564 Metasomatism vs host magma infiltration: a case study of Sal mantle xenoliths, Cape Verde
565 Archipelago. In: Coltorti, M. & Gregoire, M. (eds) *Metasomatism in Oceanic and Continental
566 Lithospheric Mantle. Geological Society of America, Special Papers* 478, 283-305.
- 567 Brey, G. P., Köhler, T., 1990. Geothermobarometry in four-phase lherzolites II. New thermobarometers
568 and practical assessment of existing thermobarometer. *Journal of Petrology* 31, 1353-1378. Cande,
569 S.C., Leslie, R.B., 1986. Late Cenozoic tectonics of the Southern Chile Trench. *Journal of Geophysical
570 Research* 91, 471-496.
- 571 Coltorti, M., Bonadiman, C., Faccini, B., Gregoire, M., O'Reilly, S. Y., Powell, W., 2007. Amphiboles
572 from suprasubduction and intraplate lithospheric mantle. *Lithos* 99, 68-84.
- 573 Corbella, H., Chelotti, L., Pomposiello, C., 1996. Neotectonica del rift Jurásico austral en Pali Aike,
574 Patagonia Extrandina, Santa Cruz, Argentina. In: XIII Congreso Geológico Argentino y III Congreso
575 de Exploración de Hidrocarburos, Actas II, 383-393.
- 576 D'Orazio M., Agostini S., Mazzarini F., Innocenti F., Manetti P., Haller M., Lahsen A., 2000. The Pali
577 Aike Volcanic Field, Patagonia: slab-window magmatism near the tip of South America.
578 *Tectonophysics* 321, 407-427.

- 579 D'Orazio M., Agostini S., Innocenti F., Haller M., Manetti P., Mazzarini F., 2001. Slab window-related
580 magmatism from southernmost South America: the Late Miocene mafic volcanics from the Estancia
581 Glencross Area (~52°S, Argentina–Chile).
- 582 Dantas, C., 2007. Caractérisation du manteau supérieur patagonien: les enclaves ultramafiques et
583 mafiques dans les laves alcalines. PhD thesis.
- 584 DeBari, S. M., Coleman, R. J., 1989. Examination of the deep levels of an island arc: evidence from the
585 Tonsina ultramafic–mafic assemblage. *Journal of Geophysical Research* 94, 4373–4391. Diraison, M.,
586 Coppold, P.R., Gapais, D., Rossello, E.A., 1997. Magellan Strait: part of a Neogene rift system.
587 *Geology* 25, 703–706.
- 588 Downes, H., 2005. Geochemical constraints on the origin of mantle pyroxenites. *American Geophysical*
589 *Union*, V41E-1505. Faccini, B., Bonadiman, C., Coltorti, C., Gregoire, M., Siena, F., 2013. Oceanic
590 Material Recycled within the Sub-Patagonian Lithospheric Mantle (Cerro del Fraile, Argentina).
591 *Journal of Petrology* 54, 1211-1258.
- 592 Foley, S. F., 2011. A reappraisal of redox melting in the Earth's mantle as a function of tectonic setting
593 and time. *Journal of Petrology* 52, 1363-1391.
- 594 Forsythe, R., Nelson, E., Carr, M.J., Keating, M.E., He & M., Mpodozis, C., Soffia, J.M. and
595 Harambour, S., 1986. Pliocene near trench magmatism in Southern Chile: a possible manifestation of
596 ridge collision. *Geology*, 14: 23-27.
- 597 Franz, L., Becker, K. P., Kramer, W., Herzig, P. M., 2002. Metasomatic mantle xenoliths from the
598 Bismark microplate (Papua New Guinea) - thermal evolution, geochemistry and extent of slab-induced
599 metasomatism. *Journal of Petrology* 43, 315-343.
- 600 Garrido, C. J., Bodinier, J. L., 1999. Diversity of mafic rocks in the Ronda peridotite: evidence for
601 pervasive melt–rock reaction during heating of subcontinental lithosphere by upwelling asthenosphere.
602 *Journal of Petrology* 40, 729-754.
- 603 Gorrington, M. L., Kay, S. M., Zeitler, P. K., Ramos, V. A., Rubiolo, D., Fernandez, M. I., Panza, J. L.,
604 1997. Neogene patagonian plateau lavas: continental magmas associated with ridge collision at the
605 Chile Triple Junction. *Tectonics* 16, 1-17.

- 606 Gorrington, M.L., Kay, S.M., 2000. Carbonatite metasomatized peridotites xenoliths from southern
607 Patagonia: implications for lithospheric processes and Neogene plateau magmatism. *Contribution to*
608 *Mineralogy and Petrology* 140, 55–72.
- 609 Gorrington, M. L., Kay, S. M., 2001. Mantle processes and sources of Neogene slab window magmas from
610 Southern Patagonia, Argentina. *Journal of Petrology* 42, 1067-1094.
- 611 Gorrington, M. L., Singer, B., Gowers, J., Kay, D. M., 2003. Plio-Pleistocene basalts from the Meseta del
612 Lago Buenos Aires, Argentina: evidence for asthenosphere-lithosphere interactions during slab window
613 magmatism. *Chemical Geology* 193, 215-235.
- 614 Green, D. H., Edgar, A. D., Beasley, P., Kiss, E., Ware, N. G., 1974. Upper mantle source for some
615 hawaiites mugearites and benmoreites. *Contribution to mineralogy and Petrology* 48, 33-44.
- 616 Hidas, K., Guzmics, T., Szabó, CS., Kovács, I., Bodnar, R. J., Zajacz, Z., Nédli, Z., Vaccari, L.,
617 Perucchi, A., 2010. Coexisting silicate melt inclusions and H₂O-bearing, CO₂-rich fluid inclusions in
618 mantle peridotite xenoliths from the Carpathian-Pannonian region (central Hungary). *Chemical*
619 *Geology* 274, 1-18.
- 620 Ionov, D. A., Hofmann, A.W., 2007. Depth of formation of subcontinental off-craton peridotites. *Earth*
621 *and Planetary Science Letters* 261, 620-634.
- 622 Ionov, D. A., Seitz, H. M., 2008. Lithium abundances and isotopic compositions in mantle xenoliths
623 from subduction and intra-plate settings: mantle sources versus eruption histories. *Earth and Planetary*
624 *Science Letters* 266, 316-331.
- 625 Johnson, K. T. M., Dick, H. J. B., Shimizu, N., 1990. Melting in the oceanic upper mantle: an ion
626 microprobe study of diopsides in abyssal peridotites. *Journal of Geophysical Research* 95, 2661-2678.
- 627 Kay, S. M., Ramos, V. A., Mpodozis, C., Sruoga, P., 1989. Late Paleozoic to Jurassic silicic magmatism
628 at the Gondwanaland margin: analogy to the Middle Proterozoic in North America? *Geology* 17, 324–
629 328.
- 630 Kay, S.M., Ramos, V.A., Marquez, M., 1993. Evidence in Cerro Pampa volcanic rocks for slab-melting
631 prior to ridge-collision in southern South America. *Journal of Geology* 101, 703- 714.

- 632 Kay, S. M., Ardolino, A., Gorrington, M. L., Ramos, V. A., 2007. The Somoncuro large igneous province in
633 Patagonia: interaction of a transient mantle thermal anomaly with a subducting slab. *Journal of*
634 *Petrology* 48, 43-77.
- 635 Kelemen, P. B., Dick, H. J. B., Quick, J. E., 1992. Formation of harzburgite by pervasive melt/rock
636 reaction in the upper mantle. *Nature* 358, 635-641.
- 637 Kempton, P.D., Hawkesworth, C.J., Lopez-Escobar, L., Pearson, D.G., Ware, A.J., 1999. Spinel±garnet
638 lherzolite xenoliths from Pali Aike: Part 2. In: Gurney, J.J., Gurney, J.L., Pascoe, M.D., Richardson,
639 S.H. (Eds.), *Trace element and isotopic evidence bearing on the evolution of lithospheric mantle*
640 *beneath southern Patagonia. The J.B. Dawson Volume. Proc. International Kimberlite Conference* 7,
641 *vol. 1*, 415–428.
- 642 Kepezhinskas, P. K., Defant, M. J., Drummond, M. S., 1995. Na metasomatism in the island-arc mantle
643 by slab melt-peridotite interaction: evidence from mantle xenoliths in the north Kamchatka arc. *Journal*
644 *of Petrology* 36, 1505-1527.
- 645 Kerr, A. C., Saunders, A. D., Tarney, A. D., Berry, N. H., Hards, V. L., 1995. Depleted mantle plume
646 geochemical signatures: no paradox for plume theories. *Geology* 23, 843-846.
- 647 Kilian, R., Franzen, C., Koch, M., 1998. The metasomatism of the mantle wedge below the southern
648 Andes: constraints from laser ablation microprobe ICP-MS trace element analysis of clinopyroxenes,
649 orthopyroxenes and fluid inclusions of mantle xenoliths. *Terra Nostra* 98/5, 81–82.
- 650 Kilian, R., Stern, C.R., 2002. Constraints on the interaction between slab melts and the mantle wedge
651 from adakitic glass in peridotite xenoliths. *European Journal of Mineralogy* 14, 25–36.
- 652 Köhler, T., Brey, G. P., 1990. Ca-exchange between olivine and clinopyroxene as a
653 geothermobarometer calibrated from 2 to 60 kbar in primitive natural lherzolites. *Geochimica et*
654 *Cosmochimica acta* 54, 2375-2388
- 655 Koloskov, A.V., Khotin, M.YU., 1978. Ultramafic inclusions in lavas of present Kamchatka volcanoes,
656 in: *Academy of Sciences of the USSR Soviet Geophysical Committee (Eds), Inclusions in the Volcanic*
657 *Rocks of the Kuril-Kamchatka Island Arc, Nauka, Moscow (in Russian with English abstract), pp. 36–*
658 *66.*

- 659 Kornprobst, J., Piboule, M., Roden, M., Tabit, A., 1990. Corundum-bearing garnet clinopyroxenites at
660 Beni Bousera (Morocco): original plagioclase-rich gabbros recrystallized at depth within mantle.
661 *Journal of Petrology* 31, 17-45.
- 662 Kurat, G., Palme, H., Spettel, B., Baddenhausen, H., Hofmeister, H., Palme, C., Wanke, H., 1980.
663 *Geochemistry of ultramafic xenoliths from Kapfenstein, Austria: evidence for a variety of upper mantle*
664 *processes. Geochimica et Cosmochimica Acta* 44, 45-60.
- 665 Lachance, G.R., Traill, R.J., 1966. Practical solution to the matrix problem in X-ray analysis. *Canadian*
666 *Spectroscopy* 11, 43-48.
- 667 Laurora, A., Mazzucchelli, M., Rivalenti, G., Vannucci, R., Zanetti, A., Barbieri, M.A., Cingolani, C.A.,
668 2001. Metasomatism and melting in carbonated peridotite xenoliths from the mantle wedge: the
669 Gobernador Gregores case (Southern Patagonia). *Journal of Petrology* 42, 69–87.
- 670 Le Bas, M. J., Streckeisen, A. L., 1991. The IUGS systematics of igneous rocks. *Journal of the*
671 *Geological Society, London*, 148, 825-833.
- 672 McDonough, W. F., Sun, S., 1995. The composition of the Earth. *Chemical Geology* 120, 223-253.
- 673 McInnes, B. I. A., Gregoire, M., Binns, R. A., Herzig, P. M., Hannington, M. D., 2001. Hydrous
674 metasomatism of oceanic subarc mantle, Lihir, Papua New Guinea: petrology and geochemistry of
675 fluid-metasomatised mantle wedge xenoliths. *Earth and Planetary Science Letters* 188, 169-183.
- 676 Mercier, J.C., Nicolas, A., 1975. Textures and fabrics of upper mantle peridotites as illustrated by
677 xenoliths from basalts. *Journal of Petrology* 16, 454– 487.
- 678 Nixon, P. H., 1987. *Mantle Xenoliths*. John Wiley & Sons, New York.
- 679 Ntaflos, T., Bjerg, E. A., Labudia, C. H., Kurat, G., 2006. Depleted lithosphere from the mantle wedge
680 beneath Tres Lagos, southern Patagonia, Argentina. *Lithos* 94, 46-65.
- 681 Pankhurst, R. J., Rapela, C. W., 1995. Production of Jurassic rhyolite by anatexis of the lower crust of
682 Patagonia. *Earth and Planetary Science Letters* 134, 223–236.
- 683 Polvé, M., Allègre, C. J., 1980. Orogenic lherzolite complexes studied by ^{87}Rb - ^{87}Sr : A clue to
684 understand the mantle convection processes? *Earth and Planetary Science Letters* 51, 71-93.

- 685 Pouchou, J. L., Pichoir, F. 1984. A new model for quantitative X-ray microanalysis. Part 1: application
686 to the analysis of homogeneous samples. *Recherche Aérospatiale* 5, 13–38.
- 687 Ramos, V.A., Kay, S.M., 1992. Southern Patagonia plateau basalts and deformation: backarc testimony
688 of ridge collisions. *Tectonophysics* 205, 261–282.
- 689 Rapela, C. W., Spalletti, L. A., Merodio, J. C., Aragon, E., 1988. Temporal evolution and spatial
690 variation of lower Tertiary volcanism in the Patagonian Andes (40°–42°30'S). *Journal of South
691 American Earth Sciences* 1, 75–88.
- 692 Rapela, C. W., Kay, S. M., 1988. Late Paleozoic to Recent magmatic evolution of northern Patagonia.
693 *Episodes* 11, 175-182.
- 694 Rivalenti, G., Mazzucchelli, M., Zanetti, A., Vannucci, R., Bollinger, C., Hemond, C., Bertotto, G. W.,
695 2007. Xenoliths from Cerro de los Chenques (Patagonia): An example of slab-related metasomatism in
696 the backarc lithospheric mantle. *Lithos* 49, 45-67
- 697 Saha, A., Basu, A. R., Jacobsen, S. B., Poreda, R. J., Yin, Q.-Z., Yogodzinski, G. M., 2005. Slab
698 devolatilization and Os and Pb mobility in the mantle wedge of the Kamchatka arc. *Earth and Planetary
699 Science Letters* 236, 182-194.
- 700 Schiano, P., Eiler, J. M., Hutcheon, I. D., Stolper, E. M., 2000. Primitive CaO-rich, silica-
701 undersaturated melts in islands arcs: evidence for the involvement of clinopyroxene-rich lithologies in
702 the petrogenesis of arc magmas. *Geochemistry, Geophysics, Geosystems* 1,
703 doi:10.1029/1999GC000032.
- 704 Smith, D., Riter, J. C. A., 1997. Genesis and evolution of low-Al orthopyroxene in spinel peridotite
705 xenoliths, Grand Canyon field, Arizona, USA. *Contributions to Mineralogy and Petrology* 127, 391-
706 404.
- 707 Smith, D., Riter, J. C. A., Mertzman, S. A., 1999. Erratum to “water-rock interactions, orthopyroxene
708 growth and Si-enrichment in the mantle: evidence in xenoliths from the Colorado Plateau, southwestern
709 United States”. *Earth and Planetary Science Letters* 167, 347-356.

- 710 Stern, C.R., Saul, S., Skewes, M.A., Futa, K., 1989. Garnet peridotites xenoliths from Pali-Aike basalts
711 of southernmost South America. Kimberlites and related rocks. Geological Society of Australia, Special
712 Publication 14, 735–744. Backwell, Carlton.
- 713 Stern, C.R., Killian, R., 1996. Role of the subducted slab, mantle wedge, and continental crust in the
714 generation of adakites from the Andean Austral Volcanic Zone. *Contribution to Mineralogy and*
715 *Petrology* 123, 263- 281.
- 716 Stern, C.R., Kilian, R., Olker, B., Hauri, E.H., Kyser, T.K., 1999. Evidence from mantle xenoliths for
717 relatively thin (<100 km) continental lithosphere below the Phanerozoic crust of southernmost South
718 America. *Lithos* 48, 217–235.
- 719 Streckeisen, A. L., 1976. Classification of the common igneous rocks by means of their chemical
720 composition: a provisional attempt. *Neues Jahrbuch für Mineralogie, Monatshefte*, 1976, H. 1, 1-15.
- 721 Sun, S., McDonough, W. F., 1989. Chemical and isotopic systematics of oceanic basalts: implications
722 for mantle composition and processes. In: *Magmatism in the ocean basins*, edited by Saunders, A. D.,
723 Norry, M. J., Geological Society of London, London, 313-345.
- 724 Takahashi, E., 1978. Petrologic model of the crust and upper mantle of the Japanese island arcs.
725 *Bulletin Volcanologique* 41, 529–47.
- 726 Thorpe, R. S., Francis, P. W., Hamill, M., Baker, M. C. W., 1982. The Andes, in *Andesite: orogenic*
727 *andesites and related rocks*, edited by R. S. Thorpe, 187-205, Chichester: Wiley.
- 728 Wang, J., Hattori, K. H., Kilian, R., Stern, C. R., 2007. Metasomatism of sub-arc mantle peridotites
729 below southernmost South America: reduction of fO_2 by slab-melt. *Contributions to Mineralogy and*
730 *Petrology* 153, 607-624.
- 731 Wang, J. H., Yin, A., Harrison, T. M., Grove, M., Yuquan, Z., Guang-Hong, X., 2001. A tectonic model
732 for Cenozoic activities in the eastern Indo-Asian collision zone. *Earth and Planetary Science Letters*
733 188, 123-133.
- 734 Weyer, S., Ionov, D. A., 2007. Partial melting and melt percolation in the mantle: The message from Fe
735 isotopes. *Earth and Planetary Science Letters* 259, 119-133.

736 Widom, E., Kepezhinskas, P., Defant, M. J., 2003. The nature of metasomatism in the sub-arc mantle
 737 wedge: evidence from Re/Os isotopes in Kamchatka peridotite xenoliths. *Chemical Geology* 196,
 738 283-306.

739 Wilkinson, J. F. G., Stolz, A. J., 1997. Subcalcic clinopyroxenites and associated ultramafic xenoliths in
 740 alkali basalt near Glenn Innes, northeastern New South Wales, Australia. *Contributions to Mineralogy
 741 and Petrology* 127, 272-290.

742 Zanetti, A., Mazzucchelli, M., Rivalenti, G., Vannucci, R., 1999. The Finero phlogopiteperidotite massif:
 743 an example of subductionrelated metasomatism. *Contributions to Mineralogy and Petrology* 134, 107–
 744 122.

745

746

747

748

749

750

751

752 **Figure caption.**

753

754 **Fig. 1:** Sketch map of Patagonia (**A**, after *D’Orazio et al., 2000*). VG stands for “Volcanic Gap”. (**a**),
 755 (**b**) and (**c**) indicate the back-arc volcanic fields respectively of Northern Patagonia, Central Patagonia
 756 and Southern Patagonia. **1** Cerro Aznare; **2** Cerro Rio Chubut; **3** Cerro de los Chenques; **4** Cerro
 757 Clark. Sketch map **B** (from *Gorring et al., 1997*) shows the occurrence of the different plateau of
 758 Central Patagonia. In grey and black are represented the main and post-plateau sequences respectively.
 759 Black star localizes sampling site of xenoliths at Estancia Sol de Mayo (ESM).

760 **Fig. 2:** Total alkali vs. silica diagram of *Le Bas and Streckeisen (1991)*. Dash dot line separates the
 761 alkaline and subalkaline domains.

762 **Fig. 3:** Chondrite normalized (*Sun and McDonough, 1989*) trace element compositions of Meseta
 763 Lago Buenos Aires (MLBA) post-plateau lavas.

764 **Fig. 4:** Ultramafic classification diagram (after *Streckeisen, 1976*) of the Estancia Sol de Mayo (ESM)
 765 mantle xenoliths. Empty symbols indicate samples studied only petrographically, while full symbols
 766 indicate those studied both petrographically and geochemically.

767 **Fig. 5:** Transmitted plane-polarized photomicrographs of representative assemblages in the Estancia
 768 Sol de Mayo (ESM) xenoliths. Ol, olivine; opx, orthopyroxene; cpx, clinopyroxene; sp, spinel. Cpx
 769 are further classified as cpx1 and cpx2. The former generally occur as protogranular in the peridotitic
 770 matrix, while the latter is observed around the sp. Opx is subdivided in opx1, opx2 and opx3: the first
 771 present as large protogranular crystals with exsolution lamellae while the second as small clean and
 772 undeformed grains without exsolution lamellae; the third occur as smaller grains arranged in vein. (A)
 773 Protogranular anhydrous spinel-bearing harzburgite MGP4b comprising ol, cpx1 and opx1. (B)
 774 Protogranular to porphyroclastic anhydrous spinel-bearing harzburgite MGP1b comprising ol, small
 775 porphyroclastic clean and undeformed grains of opx2, a cpx1 grain and a cpx2 growing around a black
 776 sp. (C) Vein of opx3 in dunite MGP2a. Opx3 are surrounded by a black matrix constituted by very
 777 small grains of ol, cpx and plagioclase.

778 **Fig. 6:** Al_2O_3 vs mg# of clinopyroxenes and orthopyroxenes. Diamonds refer to lherzolites, squares to
 779 harzburgites, triangles to dunites and asterisk to wehrlite. In A, black symbols represent cpx1 while
 780 grey symbols cpx2.

781 **Fig. 7:** Chondrite-normalized (*Sun and McDonough, 1989*) incompatible trace elements (A, B, C, D)
 782 and REE (A', B', C', D') of clinopyroxenes.

783 **Fig. 8:** Chondrite-normalized (*Sun and McDonough, 1989*) incompatible trace elements (A, B, C) and
 784 REE (A', B', C') of orthopyroxenes.

785 **Fig. 9:** Ti^* (calculated as $[Ti_N / ((Eu_N + Gd_N) / 2)]$) and Zr^* (calculated as $[Zr_N / ((Sm_N + Nd_N) / 2)]$) vs.
 786 $(Ce/Yb)_N$ for some selected orthopyroxenes from Estancia Sol de Mayo (ESM). For symbols refer to
 787 **Fig. 6.**

788 **Fig. 10:** Oxygen fugacity [calculated as $\Delta \log fO_2$ (QFM) (*Ballhaus et al., 1991*)] vs. temperature (from
 789 *Brey and Köhler, 1990*) of Estancia Sol de Mayo (ESM) peridotites. For comparison oxygen fugacity
 790 of Cerro del Fraile from *Wang et al. (2007)* and *Faccini et al. (2013)*.

791 **Fig. 11:** Chondrite-normalized REE of cpx from Estancia Sol de Mayo (ESM, white lines) from
 792 lherzolites (A) and harzburgites (B) compared to the curves (black dashed lines) of 5%, 10%, 15%,
 793 20%, 25% and 26% fractional partial melting (*Johnson et al., 1990*) of a starting fertile cpx (bold
 794 black line) from *Bonadiman et al. (2005)*.

795 **Fig. 12:** Al_2O_3 vs MgO melting trends from *Bonadiman et al., (2011)* for opx and cpx in Estancia Sol
 796 de Mayo (ESM) mantle xenoliths. Primitive Mantle (PM) opx and cpx composition in terms of Al_2O_3
 797 and MgO were calculated on the basis of the primitive mantle composition of *McDonough & Sun*
 798 (*1995*). Black crosses on curves indicate partial melting percentages.

799 **Fig. 13:** Variation in some selected samples of $(La/Yb)_N$ and Sr_N vs. Al_2O_3 of clinopyroxenes. A
 800 negative correlation between the two geochemical markers and the content of Al_2O_3 is highlighted. For
 801 symbols refer to **Fig. 6.**

802 **Fig. 14:** Chondrite-normalized (*Sun and McDonough, 1989*) incompatible trace elements (A, C, E)
 803 and REE patterns (B, D, F) of clinopyroxenes from pyroxenites of Nothern and Central Patagonia and
 804 wehrlites from Central Patagonia (*Dantas, 2007*). Light grey field represents the clinopyroxenes from
 805 Estancia Sol de Mayo (ESM).

806 **Fig. 15:** Chondrite-normalized (*Sun and McDonough, 1989*) incompatible trace elements of
 807 clinopyroxenes from pyroxenites of Nothern and Central Patagonia (*Dantas, 2007*) and

808 clinopyroxenes phenocrysts entrained in Northern Ethiopian continental flood basalts (from *Beccaluva*
809 *et al.*, 2004).

810 **Fig. 16:** Chondrite-normalized (*Sun and McDonough, 1989*) incompatible trace elements pattern
811 reconstructions of clinopyroxenes from transitional/alkaline mafic lavas from the Triple Junction
812 Province.

813

814 **Table caption.**

815 **Table 1:** Bulk rock major (in wt. %) and trace (in ppm) element analysis of six host lavas from
816 Estancia Sol de Mayo (ESM). Mg# ($\text{MgO}/(\text{MgO}+\text{FeO})$ mol %) is calculated with $\text{Fe}_2\text{O}_3=0.15*\text{FeO}$
817 (*Green et al.*, 1974).

818 **Table 2:** Representative major element composition (in wt. %) of olivines of Estancia Sol de Mayo
819 (ESM) mantle xenoliths. Ol: olivine; Fo: forsterite.

820 **Table 3:** Representative major element composition (in wt. %) of clinopyroxenes of Estancia Sol de
821 Mayo (ESM) mantle xenoliths.

822 **Table 4:** Representative major element composition (in wt. %) of orthopyroxenes of Estancia Sol de
823 Mayo (ESM) mantle xenoliths.

824 **Table 5:** Representative major element composition (in wt. %) of spinels of Estancia Sol de Mayo
825 (ESM) mantle xenoliths.

826 **Table 6:** Trace element contents (ppm) of Estancia Sol de Mayo (ESM) clinopyroxenes.

827 **Table 7:** Trace element contents (ppm) of Estancia Sol de Mayo (ESM) orthopyroxenes.

828 **Table 8:** Equilibration temperature, pressure and $f\text{O}_2$ estimates of Estancia Sol de Mayo (ESM)
829 mantle xenoliths.

Table 1: Bulk rock major (in wt. %) and trace element (in ppm) analysis of six host lavas from Estancia Sol de Mayo (ESM).

| Sample | MGP1 | MGP2 | MGP3 | MGP4 | MGP5 | MGP6 |
|------------------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|
| Rock type | Basaltic trachy-andesite | Basaltic trachy-andesite | Basaltic trachy-andesite | Basaltic trachy-andesite | Basaltic trachy-andesite | Basaltic trachy-andesite |
| SiO ₂ (wt. %) | 50.80 | 50.50 | 50.60 | 50.70 | 50.50 | 50.80 |
| TiO ₂ | 2.13 | 2.15 | 2.14 | 2.10 | 2.13 | 2.14 |
| Al ₂ O ₃ | 16.30 | 16.30 | 16.40 | 16.50 | 16.40 | 16.20 |
| Fe ₂ O ₃ Tot | 9.28 | 9.57 | 9.43 | 9.31 | 9.52 | 9.49 |
| MnO | 0.15 | 0.16 | 0.16 | 0.16 | 0.16 | 0.15 |
| MgO | 3.93 | 4.02 | 4.13 | 4.00 | 4.11 | 4.04 |
| CaO | 7.52 | 7.62 | 7.48 | 7.46 | 7.45 | 7.42 |
| Na ₂ O | 5.69 | 5.59 | 5.45 | 5.64 | 5.57 | 5.63 |
| K ₂ O | 2.65 | 2.58 | 2.51 | 2.55 | 2.57 | 2.55 |
| P ₂ O ₅ | 1.09 | 1.08 | 1.07 | 1.07 | 1.06 | 1.07 |
| LOI | 0.35 | 0.36 | 0.56 | 0.43 | 0.45 | 0.40 |
| Total | 99.89 | 99.93 | 99.93 | 99.92 | 99.92 | 99.89 |
| mg# | 48.65 | 48.45 | 49.51 | 49.05 | 49.15 | 48.79 |
| Ni (ppm) | 45.6 | 44.7 | 46.6 | 46.8 | 47.3 | 44.9 |
| Co | 24.1 | 24.1 | 25.5 | 26.2 | 25.0 | 27.1 |
| Cr | 195 | 182 | 210 | 173 | 174 | 164 |
| V | 157 | 157 | 161 | 155 | 158 | 158 |
| Sc | 19.6 | 19.9 | 20.5 | 18.6 | 20.8 | 18.6 |
| Sr | 682 | 689 | 681 | 674 | 685 | 678 |
| Rb | 32.4 | 32.2 | 31.2 | 31.3 | 32.1 | 32.3 |
| Ba | 398 | 393 | 376 | 390 | 387 | 387 |
| Zr | 256 | 256 | 253 | 251 | 254 | 252 |
| Nb | 49.5 | 49.4 | 49.6 | 47.2 | 49.8 | 48.2 |
| Th | 4.63 | 4.24 | 5.22 | 4.66 | 5.75 | 3.15 |
| Y | 20.4 | 21.3 | 20.5 | 20.6 | 21.4 | 20.5 |
| La | 26.9 | 26.9 | 29.0 | 29.3 | 25.8 | 25.5 |
| Ce | 91.8 | 91.7 | 96.6 | 98.1 | 88.4 | 78.6 |
| Nd | 37.8 | 38.5 | 36.3 | 37.9 | 36.5 | 38.2 |
| Pb | 9.44 | 9.60 | 6.58 | 10.8 | 12.0 | 12.3 |
| Zn | 66.4 | 70.2 | 67.3 | 66.3 | 66.6 | 67.5 |
| Cu | 39.7 | 38.3 | 40.0 | 38.9 | 38.9 | 40.4 |
| Ga | 30.3 | 35.2 | 28.6 | 27.5 | 28.0 | 27.7 |

Mg# ((MgO/(MgO+FeO) mol %) is calculated with Fe₂O₃=0.15*FeO (*Green et al., 1974*)

Table 2: Representative major element composition (in wt. %) of olivines of Estancia Sol de Mayo (ESM) mantle xenoliths.

| Sample phase | MGP2b | | | MGP2b2 | | | MGP1b | | | MGP1g | | | MGP3b | | | MGP4b | | | MGP1h | | | MGP2a | | | MGP1d | | | | |
|--------------------------------|--------------|--------|--------|--------------|--------|--------------|-------------|--------|--------|-------------|--------|-------------|-------------|--------|-------|-------------|-------|-------------|--------|--------|-------|--------|-------|--------|------------|--------|----|-----|------|
| | ol | ol | ol | rim | core | Lherzolitite | ol | ol | ol | rim | core | Harzburgite | ol | ol | ol | rim | core | Harzburgite | ol | ol | ol | rim | core | Dunite | ol | ol | ol | rim | core |
| Host rock | Lherzolitite | | | Lherzolitite | | | Harzburgite | | | Harzburgite | | | Harzburgite | | | Harzburgite | | | Dunite | | | Dunite | | | Wehrlitite | | | | |
| SiO ₂ | 40.60 | 41.36 | 41.30 | 41.32 | 41.30 | 41.32 | 40.39 | 40.01 | 40.05 | 39.22 | 40.83 | 40.89 | 41.00 | 41.25 | 40.73 | 40.44 | 39.87 | 39.63 | 41.00 | 41.25 | 40.73 | 40.44 | 39.87 | 39.63 | 39.35 | 38.86 | | | |
| FeO | 8.74 | 9.11 | 8.83 | 9.08 | 8.83 | 9.08 | 11.25 | 11.60 | 14.14 | 15.01 | 10.60 | 9.74 | 8.13 | 8.20 | 9.03 | 8.96 | 9.76 | 10.32 | 8.13 | 8.20 | 9.03 | 8.96 | 9.76 | 10.32 | 16.96 | 17.40 | | | |
| MnO | 0.21 | 0.25 | 0.15 | 0.17 | 0.22 | 0.17 | 0.22 | 0.17 | 0.16 | 0.20 | 0.20 | 0.12 | 0.12 | 0.15 | 0.10 | 0.12 | 0.14 | 0.24 | 0.12 | 0.15 | 0.10 | 0.12 | 0.14 | 0.24 | 0.25 | 0.20 | | | |
| MgO | 50.26 | 50.20 | 49.34 | 49.86 | 47.68 | 48.08 | 0.09 | 0.09 | 46.15 | 45.43 | 49.13 | 49.36 | 50.87 | 50.39 | 49.14 | 49.07 | 48.94 | 49.20 | 50.87 | 50.39 | 49.14 | 49.07 | 48.94 | 49.20 | 43.66 | 43.62 | | | |
| CaO | 0.07 | 0.04 | 0.01 | 0.06 | 0.34 | 0.46 | 0.34 | 0.46 | 0.13 | 0.04 | 0.08 | 0.04 | <0.01 | 0.15 | 0.07 | <0.01 | 0.08 | 0.05 | <0.01 | 0.15 | 0.07 | <0.01 | 0.08 | 0.05 | 0.07 | 0.07 | | | |
| NiO | 0.50 | 0.27 | 0.46 | 0.36 | 0.12 | 0.11 | 0.12 | 0.11 | 0.25 | 0.23 | 0.35 | <0.01 | 0.41 | <0.01 | 0.34 | 0.37 | 0.21 | 0.45 | 0.41 | <0.01 | 0.34 | 0.37 | 0.21 | 0.45 | 0.15 | 0.23 | | | |
| Cr ₂ O ₃ | <0.01 | 0.09 | 0.02 | 0.01 | 0.12 | 0.11 | 0.12 | 0.11 | 0.06 | <0.01 | 0.07 | 0.08 | 0.13 | 0.18 | 0.02 | <0.01 | <0.01 | 0.03 | 0.13 | 0.18 | 0.02 | <0.01 | <0.01 | 0.03 | 0.02 | 0.02 | | | |
| Total | 100.38 | 101.32 | 100.11 | 100.86 | 100.09 | 100.52 | 100.09 | 100.52 | 100.94 | 100.13 | 101.26 | 100.23 | 100.66 | 100.32 | 99.43 | 98.96 | 99.00 | 99.92 | 100.66 | 100.32 | 99.43 | 98.96 | 99.00 | 99.92 | 100.46 | 100.40 | | | |
| Fo | 91.11 | 90.76 | 90.88 | 90.73 | 88.31 | 88.08 | 88.31 | 88.08 | 85.33 | 84.36 | 89.20 | 90.03 | 91.77 | 91.63 | 90.65 | 90.71 | 89.94 | 89.47 | 91.77 | 91.63 | 90.65 | 90.71 | 89.94 | 89.47 | 82.10 | 81.71 | | | |
| Si | 1.003 | 1.022 | 1.020 | 1.021 | 0.998 | 0.988 | 0.998 | 0.988 | 0.989 | 0.969 | 1.009 | 1.010 | 1.013 | 1.019 | 1.006 | 0.999 | 0.985 | 0.979 | 1.013 | 1.019 | 1.006 | 0.999 | 0.985 | 0.979 | 0.972 | 0.960 | | | |
| Fe ²⁺ | 0.181 | 0.188 | 0.182 | 0.188 | 0.232 | 0.240 | 0.232 | 0.240 | 0.292 | 0.310 | 0.219 | 0.201 | 0.168 | 0.169 | 0.187 | 0.185 | 0.202 | 0.213 | 0.168 | 0.169 | 0.187 | 0.185 | 0.202 | 0.213 | 0.350 | 0.359 | | | |
| Mn | 0.004 | 0.005 | 0.003 | 0.004 | 0.005 | 0.004 | 0.005 | 0.004 | 0.003 | 0.004 | 0.004 | 0.003 | 0.003 | 0.003 | 0.002 | 0.003 | 0.003 | 0.005 | 0.003 | 0.003 | 0.002 | 0.003 | 0.003 | 0.005 | 0.005 | 0.004 | | | |
| Mg | 1.851 | 1.849 | 1.817 | 1.836 | 1.756 | 1.770 | 1.756 | 1.770 | 1.699 | 1.673 | 1.809 | 1.818 | 1.873 | 1.856 | 1.809 | 1.807 | 1.802 | 1.812 | 1.873 | 1.856 | 1.809 | 1.807 | 1.802 | 1.812 | 1.608 | 1.606 | | | |
| Ca | 0.002 | 0.001 | 0.000 | 0.002 | 0.002 | 0.002 | 0.002 | 0.002 | 0.003 | 0.001 | 0.002 | 0.001 | 0.000 | 0.004 | 0.002 | 0.000 | 0.002 | 0.001 | 0.000 | 0.004 | 0.002 | 0.000 | 0.002 | 0.001 | 0.002 | 0.002 | | | |
| Ni | 0.010 | 0.005 | 0.009 | 0.007 | 0.007 | 0.009 | 0.007 | 0.009 | 0.005 | 0.005 | 0.007 | 0.000 | 0.008 | 0.000 | 0.007 | 0.007 | 0.004 | 0.009 | 0.008 | 0.000 | 0.007 | 0.007 | 0.004 | 0.009 | 0.003 | 0.005 | | | |
| Cr | 0.000 | 0.001 | 0.000 | 0.000 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.000 | 0.001 | 0.001 | 0.001 | 0.002 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.002 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | | | |
| Sum cat | 3.050 | 3.071 | 3.032 | 3.057 | 3.001 | 3.015 | 3.001 | 3.015 | 2.993 | 2.962 | 3.051 | 3.033 | 3.066 | 3.054 | 3.013 | 3.001 | 2.999 | 3.020 | 3.066 | 3.054 | 3.013 | 3.001 | 2.999 | 3.020 | 2.941 | 2.937 | | | |

Ol: olivine; Fo: forsterite.

Table 3: Representative major element composition (in wt. %) of clinopyroxenes of Estancia Sol de Mayo (ESM).

| Sample phase | MGP2b | | | MGP1b | | | MGP1c | | | MGP3b | | | MGP4b | | | MGP1h | | | MGP2a | | | MGP1d | | | | |
|--------------------------------|------------|-----------|-------|------------|------------|-------------|------------|------------|-------------|-----------|------------|-------------|------------|------------|-------------|------------|------------|--------|------------|------------|--------|------------|------------|----------|-------|--------|
| | cpx 1 core | cpx 2 rim | lherz | cpx 1 core | cpx 2 core | Harzburgite | cpx 2 core | cpx 2 core | Harzburgite | cpx 1 rim | cpx 1 core | Harzburgite | cpx 1 core | cpx 2 core | Harzburgite | cpx 2 core | cpx 2 core | Dumite | cpx 1 core | cpx 2 core | Dumite | cpx 1 core | cpx 2 core | Wehrlite | | |
| Host rock | | | | | | | | | | | | | | | | | | | | | | | | | | |
| SiO ₂ | 52.43 | 52.33 | 51.93 | 52.78 | 52.90 | 52.15 | 52.45 | 52.25 | 53.47 | 53.48 | 53.05 | 52.62 | 51.83 | 52.77 | 53.27 | 52.85 | 49.91 | 49.44 | 53.27 | 52.85 | 51.83 | 52.77 | 53.27 | 52.85 | 49.91 | 49.44 |
| TiO ₂ | 0.17 | 0.19 | 0.16 | 0.28 | 0.13 | 0.17 | 0.28 | 0.31 | 0.10 | 0.11 | 0.25 | 0.25 | 0.24 | 0.24 | 0.07 | 0.30 | 1.01 | 1.17 | 0.07 | 0.30 | 0.24 | 0.24 | 0.07 | 0.30 | 1.01 | 1.17 |
| Al ₂ O ₃ | 4.06 | 3.96 | 3.22 | 4.04 | 3.10 | 3.56 | 3.46 | 4.13 | 2.34 | 2.69 | 4.00 | 4.20 | 3.78 | 3.76 | 1.93 | 2.65 | 6.50 | 7.30 | 1.93 | 2.65 | 3.78 | 3.76 | 1.93 | 2.65 | 6.50 | 7.30 |
| Cr ₂ O ₃ | 1.03 | 1.45 | 1.20 | 1.21 | 0.92 | 0.96 | 1.32 | 1.38 | 1.06 | 1.45 | 1.28 | 1.36 | 1.25 | 1.33 | 0.87 | 1.04 | 0.60 | 0.71 | 0.87 | 1.04 | 1.25 | 1.33 | 0.87 | 1.04 | 0.60 | 0.71 |
| FeO | 2.91 | 2.9 | 2.94 | 2.99 | 3.75 | 3.69 | 3.07 | 3.07 | 3.03 | 2.98 | 2.75 | 2.80 | 3.13 | 3.20 | 3.05 | 3.16 | 4.98 | 5.07 | 3.05 | 3.16 | 3.13 | 3.20 | 3.05 | 3.16 | 4.98 | 5.07 |
| MnO | 0.15 | <0.01 | 0.15 | 0.11 | 0.16 | 0.16 | 0.10 | 0.07 | 0.09 | 0.03 | 0.10 | 0.05 | 0.10 | 0.06 | 0.09 | 0.14 | 0.15 | 0.20 | 0.09 | 0.14 | 0.10 | 0.06 | 0.09 | 0.14 | 0.15 | 0.20 |
| MgO | 16.68 | 16.60 | 17.16 | 16.80 | 16.82 | 16.70 | 16.49 | 16.34 | 17.24 | 17.01 | 16.99 | 17.14 | 16.49 | 16.55 | 17.45 | 17.11 | 14.69 | 14.55 | 17.45 | 17.11 | 16.49 | 16.55 | 17.45 | 17.11 | 14.69 | 14.55 |
| CaO | 20.95 | 21.59 | 21.24 | 21.05 | 21.12 | 20.98 | 21.29 | 21.05 | 21.32 | 21.04 | 21.15 | 21.02 | 20.76 | 21.07 | 21.21 | 20.49 | 20.57 | 20.60 | 21.21 | 20.49 | 20.76 | 21.07 | 21.21 | 20.49 | 20.57 | 20.60 |
| Na ₂ O | 0.81 | 0.99 | 0.81 | 0.91 | 0.81 | 0.78 | 1.05 | 0.95 | 0.92 | 0.92 | 1.09 | 1.03 | 1.04 | 1.06 | 0.95 | 1.18 | 0.96 | 1.00 | 0.95 | 1.18 | 1.04 | 1.06 | 0.95 | 1.18 | 0.96 | 1.00 |
| Total | 99.19 | 99.97 | 98.81 | 100.17 | 99.71 | 99.15 | 99.51 | 99.55 | 99.57 | 99.71 | 100.66 | 100.47 | 98.62 | 100.04 | 98.89 | 98.92 | 99.37 | 100.04 | 98.89 | 98.92 | 98.62 | 100.04 | 98.89 | 98.92 | 99.37 | 100.04 |
| mg# | 91.1 | 91.2 | 91.2 | 90.9 | 88.9 | 89.0 | 90.5 | 90.5 | 91.0 | 91.1 | 91.7 | 91.6 | 90.4 | 90.2 | 91.1 | 90.6 | 84.0 | 83.6 | 91.1 | 90.6 | 90.4 | 90.2 | 91.1 | 90.6 | 84.0 | 83.6 |
| Si | 1.913 | 1.895 | 1.901 | 1.907 | 1.925 | 1.907 | 1.910 | 1.903 | 1.943 | 1.944 | 1.904 | 1.891 | 1.902 | 1.910 | 1.946 | 1.930 | 1.833 | 1.804 | 1.946 | 1.930 | 1.902 | 1.910 | 1.946 | 1.930 | 1.833 | 1.804 |
| Ti | 0.005 | 0.005 | 0.004 | 0.008 | 0.004 | 0.005 | 0.008 | 0.008 | 0.003 | 0.003 | 0.007 | 0.007 | 0.007 | 0.007 | 0.002 | 0.008 | 0.028 | 0.032 | 0.002 | 0.008 | 0.007 | 0.007 | 0.002 | 0.008 | 0.028 | 0.032 |
| Al | 0.175 | 0.169 | 0.139 | 0.172 | 0.133 | 0.153 | 0.148 | 0.177 | 0.100 | 0.115 | 0.169 | 0.178 | 0.164 | 0.160 | 0.083 | 0.114 | 0.281 | 0.314 | 0.083 | 0.114 | 0.164 | 0.160 | 0.083 | 0.114 | 0.281 | 0.314 |
| Fe ³⁺ | 0.004 | 0.015 | 0.019 | 0.007 | 0.010 | 0.013 | 0.013 | 0.007 | 0.011 | 0.004 | 0.012 | 0.015 | 0.014 | 0.011 | 0.016 | 0.016 | 0.012 | 0.017 | 0.016 | 0.016 | 0.014 | 0.011 | 0.016 | 0.016 | 0.012 | 0.017 |
| Fe ²⁺ | 0.085 | 0.072 | 0.071 | 0.083 | 0.104 | 0.100 | 0.081 | 0.087 | 0.081 | 0.087 | 0.070 | 0.069 | 0.082 | 0.086 | 0.077 | 0.081 | 0.141 | 0.138 | 0.077 | 0.081 | 0.082 | 0.086 | 0.077 | 0.081 | 0.141 | 0.138 |
| Mn | 0.005 | 0.000 | 0.005 | 0.003 | 0.005 | 0.005 | 0.003 | 0.002 | 0.003 | 0.001 | 0.003 | 0.002 | 0.003 | 0.002 | 0.003 | 0.004 | 0.005 | 0.006 | 0.003 | 0.004 | 0.003 | 0.002 | 0.003 | 0.004 | 0.005 | 0.006 |
| Mg | 0.907 | 0.896 | 0.936 | 0.905 | 0.912 | 0.910 | 0.895 | 0.887 | 0.934 | 0.921 | 0.909 | 0.918 | 0.902 | 0.893 | 0.950 | 0.931 | 0.804 | 0.791 | 0.950 | 0.931 | 0.902 | 0.893 | 0.950 | 0.931 | 0.804 | 0.791 |
| Ca | 0.819 | 0.838 | 0.833 | 0.815 | 0.823 | 0.822 | 0.830 | 0.822 | 0.830 | 0.819 | 0.813 | 0.810 | 0.816 | 0.817 | 0.830 | 0.802 | 0.810 | 0.805 | 0.830 | 0.802 | 0.816 | 0.817 | 0.830 | 0.802 | 0.810 | 0.805 |
| Na | 0.057 | 0.070 | 0.057 | 0.064 | 0.057 | 0.055 | 0.074 | 0.067 | 0.065 | 0.065 | 0.076 | 0.072 | 0.074 | 0.074 | 0.067 | 0.084 | 0.068 | 0.071 | 0.067 | 0.084 | 0.074 | 0.074 | 0.067 | 0.084 | 0.068 | 0.071 |
| Cr | 0.030 | 0.042 | 0.035 | 0.035 | 0.026 | 0.028 | 0.038 | 0.040 | 0.030 | 0.042 | 0.036 | 0.039 | 0.036 | 0.038 | 0.025 | 0.030 | 0.017 | 0.020 | 0.025 | 0.030 | 0.036 | 0.038 | 0.025 | 0.030 | 0.017 | 0.020 |
| Sum cat | 4.000 | 4.002 | 4.000 | 3.999 | 3.999 | 3.998 | 4.000 | 4.000 | 4.000 | 4.001 | 3.999 | 4.001 | 4.000 | 3.998 | 3.999 | 4.000 | 3.999 | 3.998 | 3.999 | 4.000 | 4.000 | 3.998 | 3.999 | 4.000 | 3.999 | 3.998 |

All Fe as Fe⁺²; mg# = 100 x [Mg/(Mg + Fe)]; Fe⁺² and Fe⁺³ calculated by stoichiometry of the formula unit.

Table 4: Representative major element composition (in wt. %) of orthopyroxenes of Estancia Sol de Mayo (ESM).

| Sample phase | MGP2b | | MGP2b2 | | MGP1b | | MGP1c | | MGP1g | | MGP3b | | MGP4b | | MGP2a | |
|--------------------------------|-------------|-----------|-------------|-----------|-------------|----------|-------------|----------|-------------|-----------|-------------|-----------|-------------|-----------|----------|-----------|
| | opx1 rim | opx2 core | opx1 rim | opx2 core | opx1 core | opx2 rim | opx1 core | opx2 rim | opx1 core | opx2 core | opx1 rim | opx2 core | opx1 rim | opx2 core | opx3 rim | opx3 core |
| Host rock | Lherzollite | | Lherzollite | | Harzburgite | | Harzburgite | | Harzburgite | | Harzburgite | | Harzburgite | | Dunite | |
| SiO ₂ | 55.61 | 56.33 | 56.12 | 55.24 | 55.79 | 55.55 | 56.75 | 56.22 | 54.43 | 54.87 | 56.78 | 56.67 | 56.25 | 56.30 | 55.10 | 55.00 |
| TiO ₂ | 0.03 | 0.08 | 0.06 | 0.02 | <0.01 | 0.04 | <0.01 | 0.08 | 0.21 | 0.11 | 0.05 | 0.02 | 0.08 | 0.06 | 0.04 | 0.01 |
| Al ₂ O ₃ | 2.75 | 2.67 | 2.84 | 2.76 | 2.30 | 2.61 | 1.39 | 1.83 | 3.10 | 2.98 | 1.69 | 1.57 | 2.77 | 2.67 | 3.37 | 3.27 |
| FeOtot | 5.82 | 5.81 | 5.65 | 6.02 | 7.97 | 7.62 | 5.67 | 6.31 | 9.31 | 9.05 | 5.90 | 6.01 | 5.16 | 5.33 | 6.48 | 6.21 |
| MnO | 0.14 | 0.17 | 0.24 | 0.16 | 0.07 | 0.19 | 0.24 | 0.15 | 0.17 | 0.26 | 0.17 | 0.16 | 0.14 | 0.09 | 0.16 | 0.14 |
| MgO | 33.89 | 33.91 | 33.86 | 33.92 | 33.05 | 32.84 | 34.79 | 34.30 | 31.50 | 31.69 | 34.14 | 34.31 | 34.58 | 34.62 | 33.31 | 33.39 |
| CaO | 0.94 | 0.97 | 0.89 | 0.99 | 0.96 | 0.90 | 0.72 | 0.91 | 0.87 | 0.94 | 0.94 | 0.96 | 0.87 | 0.86 | 0.83 | 0.85 |
| Na ₂ O | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 | <0.01 |
| Cr ₂ O ₃ | 0.70 | 0.52 | 0.51 | 0.57 | 0.58 | 0.54 | 0.26 | 0.51 | 0.38 | 0.42 | 0.53 | 0.56 | 0.70 | 0.56 | 0.33 | 0.22 |
| Tot | 99.88 | 100.46 | 100.17 | 99.68 | 100.72 | 100.29 | 99.82 | 100.31 | 99.97 | 100.32 | 100.20 | 100.26 | 100.55 | 100.49 | 99.62 | 99.09 |
| mg# | 91.2 | 91.2 | 91.4 | 90.9 | 88.1 | 88.5 | 91.6 | 90.6 | 85.8 | 86.2 | 91.2 | 91.1 | 92.3 | 92.1 | 90.2 | 90.6 |
| Si | 1.919 | 1.934 | 1.931 | 1.908 | 1.924 | 1.923 | 1.950 | 1.932 | 1.904 | 1.912 | 1.955 | 1.949 | 1.922 | 1.925 | 1.908 | 1.913 |
| Ti | 0.001 | 0.002 | 0.002 | 0.001 | 0.000 | 0.001 | 0.000 | 0.002 | 0.006 | 0.003 | 0.001 | 0.001 | 0.002 | 0.002 | 0.001 | 0.000 |
| Al | 0.112 | 0.108 | 0.115 | 0.112 | 0.094 | 0.107 | 0.056 | 0.074 | 0.128 | 0.122 | 0.069 | 0.064 | 0.112 | 0.108 | 0.138 | 0.134 |
| Fe ³⁺ | 0.007 | 0.002 | 0.002 | 0.014 | 0.010 | 0.007 | 0.009 | 0.011 | 0.011 | 0.009 | 0.001 | 0.005 | 0.005 | 0.006 | 0.009 | 0.008 |
| Fe ²⁺ | 0.160 | 0.165 | 0.161 | 0.160 | 0.219 | 0.213 | 0.163 | 0.170 | 0.261 | 0.254 | 0.169 | 0.167 | 0.142 | 0.146 | 0.179 | 0.172 |
| Mn | 0.004 | 0.005 | 0.007 | 0.005 | 0.002 | 0.006 | 0.007 | 0.004 | 0.005 | 0.008 | 0.005 | 0.005 | 0.004 | 0.003 | 0.005 | 0.004 |
| Mg | 1.743 | 1.735 | 1.736 | 1.746 | 1.699 | 1.695 | 1.781 | 1.757 | 1.642 | 1.645 | 1.751 | 1.759 | 1.761 | 1.764 | 1.719 | 1.731 |
| Ca | 0.035 | 0.036 | 0.033 | 0.037 | 0.035 | 0.033 | 0.026 | 0.034 | 0.033 | 0.035 | 0.035 | 0.035 | 0.032 | 0.032 | 0.031 | 0.032 |
| Na | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Cr | 0.019 | 0.014 | 0.014 | 0.016 | 0.016 | 0.015 | 0.007 | 0.014 | 0.010 | 0.012 | 0.014 | 0.015 | 0.019 | 0.015 | 0.009 | 0.006 |
| Sum cat | 4.000 | 4.001 | 4.001 | 3.999 | 3.999 | 4.000 | 3.999 | 3.998 | 4.000 | 4.000 | 4.000 | 4.000 | 3.999 | 4.001 | 3.999 | 4.000 |

All Fe as Fe⁺², mg# = 100 x [Mg/(Mg + Fe)]; Fe⁺² and Fe⁺³ calculated by stoichiometry of the formula unit.

Table 5: Representative major element composition (in wt. %) of spinels of Estancia Sol de Mayo (ESM).

| Sample phase | MGP2b | | MGP2b2 | | MGP1b | | MGP1c | | MGP1g | | MGP3b | | MGP4b | | MGP1h | | MGP2a | | MGP1d | |
|--------------------------------|--------------------------|---------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|---------------------------|--------------------------|
| | SP _{epx} rim | SP _{epx} core | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim | SP _{epx} core | SP _{epx} rim |
| Host rock | Lherzollite | Lherzollite | Lherzollite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Harzburgite | Dumite | Dumite | Dumite | Dumite | Wehrllite | Wehrllite |
| SiO ₂ | 0.04 | 0.06 | 0.04 | 0.01 | 0.07 | <0.01 | 0.09 | 0.11 | 0.09 | 0.06 | 0.09 | 0.06 | 0.03 | 0.07 | 0.03 | 0.04 | 0.03 | 0.03 | 0.02 | <0.01 |
| TiO ₂ | 0.27 | 0.43 | 0.24 | 0.19 | 0.43 | 0.42 | 1.70 | 2.05 | 0.42 | 0.35 | 0.42 | 0.35 | 0.20 | 0.13 | 0.36 | 0.39 | 0.73 | 0.53 | 0.49 | 0.38 |
| Al ₂ O ₃ | 34.91 | 35.86 | 35.36 | 34.89 | 27.32 | 26.29 | 34.90 | 35.02 | 19.96 | 19.71 | 19.96 | 19.71 | 33.45 | 33.72 | 29.25 | 22.83 | 25.06 | 21.59 | 45.40 | 48.00 |
| Cr ₂ O ₃ | 32.69 | 31.46 | 32.20 | 32.07 | 36.67 | 37.96 | 18.75 | 18.27 | 43.93 | 43.88 | 43.93 | 43.88 | 33.89 | 33.08 | 34.09 | 41.43 | 38.81 | 42.20 | 17.79 | 15.20 |
| Fe ₂ O ₃ | 2.20 | 2.62 | 3.24 | 3.69 | 6.50 | 6.69 | 13.15 | 12.24 | 7.46 | 7.39 | 7.46 | 7.39 | 3.63 | 3.73 | 6.58 | 7.33 | 5.12 | 6.40 | 4.26 | 4.65 |
| FeO | 11.68 | 11.43 | 11.55 | 11.10 | 13.00 | 13.36 | 17.59 | 19.09 | 13.59 | 13.81 | 13.59 | 13.81 | 10.40 | 9.96 | 12.18 | 13.35 | 13.95 | 14.37 | 16.01 | 15.28 |
| MnO | 0.23 | 0.22 | 0.24 | 0.19 | 0.26 | 0.14 | 0.23 | 0.16 | 0.22 | 0.15 | 0.22 | 0.15 | 0.10 | 0.19 | 0.14 | 0.27 | 0.16 | 0.20 | 0.17 | 0.10 |
| MgO | 16.67 | 17.04 | 16.90 | 16.95 | 15.14 | 14.87 | 13.80 | 13.07 | 14.15 | 13.79 | 14.15 | 13.79 | 17.35 | 17.50 | 15.61 | 14.56 | 14.27 | 13.68 | 15.06 | 15.83 |
| NiO | <0.01 | 0.23 | 0.22 | 0.29 | 0.21 | <0.25 | 0.24 | 0.24 | 0.14 | 0.20 | 0.14 | 0.20 | 0.25 | 0.15 | 0.28 | 0.21 | 0.16 | 0.12 | 0.24 | 0.21 |
| Tot | 98.69 | 99.35 | 99.99 | 99.38 | 99.60 | 99.98 | 100.45 | 100.25 | 99.96 | 99.34 | 99.96 | 99.34 | 99.30 | 98.53 | 98.52 | 100.41 | 98.29 | 99.12 | 99.44 | 99.65 |
| mg# | 71.8 | 72.7 | 72.3 | 73.1 | 67.5 | 66.5 | 58.3 | 55.0 | 65.0 | 64.0 | 65.0 | 64.0 | 74.8 | 75.8 | 69.5 | 66.0 | 64.6 | 62.9 | 62.6 | 64.9 |
| cr# | 38.6 | 37.1 | 37.9 | 38.1 | 42.6 | 42.8 | 26.5 | 25.9 | 59.6 | 59.9 | 59.6 | 59.9 | 40.5 | 39.7 | 43.9 | 54.9 | 51.0 | 56.7 | 20.8 | 17.5 |
| Si | 0.001 | 0.002 | 0.001 | 0.000 | 0.002 | 0.000 | 0.003 | 0.003 | 0.003 | 0.002 | 0.003 | 0.002 | 0.001 | 0.002 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.000 |
| Ti | 0.006 | 0.009 | 0.005 | 0.004 | 0.010 | 0.009 | 0.037 | 0.045 | 0.010 | 0.008 | 0.010 | 0.008 | 0.004 | 0.003 | 0.008 | 0.009 | 0.017 | 0.012 | 0.010 | 0.008 |
| Al | 1.190 | 1.210 | 1.191 | 1.182 | 0.963 | 0.930 | 1.199 | 1.210 | 0.727 | 0.725 | 0.727 | 0.725 | 1.138 | 1.151 | 1.030 | 0.817 | 0.906 | 0.789 | 1.496 | 1.557 |
| Fe ³⁺ | 0.050 | 0.060 | 0.070 | 0.080 | 0.150 | 0.150 | 0.290 | 0.270 | 0.170 | 0.170 | 0.170 | 0.170 | 0.080 | 0.080 | 0.150 | 0.170 | 0.120 | 0.150 | 0.090 | 0.100 |
| Fe ²⁺ | 0.280 | 0.270 | 0.280 | 0.270 | 0.330 | 0.340 | 0.430 | 0.470 | 0.350 | 0.360 | 0.350 | 0.360 | 0.250 | 0.240 | 0.300 | 0.340 | 0.360 | 0.370 | 0.370 | 0.350 |
| Mn | 0.006 | 0.005 | 0.006 | 0.005 | 0.007 | 0.004 | 0.006 | 0.004 | 0.006 | 0.004 | 0.006 | 0.004 | 0.002 | 0.005 | 0.004 | 0.007 | 0.004 | 0.005 | 0.004 | 0.002 |
| Mg | 0.719 | 0.727 | 0.719 | 0.726 | 0.675 | 0.665 | 0.600 | 0.571 | 0.652 | 0.641 | 0.652 | 0.641 | 0.746 | 0.755 | 0.695 | 0.659 | 0.652 | 0.632 | 0.627 | 0.649 |
| Cr | 0.748 | 0.712 | 0.727 | 0.729 | 0.867 | 0.900 | 0.432 | 0.423 | 1.074 | 1.082 | 1.074 | 1.082 | 0.773 | 0.758 | 0.805 | 0.995 | 0.941 | 1.035 | 0.393 | 0.331 |
| Ni | 0.000 | 0.005 | 0.005 | 0.007 | 0.005 | 0.006 | 0.006 | 0.006 | 0.003 | 0.005 | 0.003 | 0.005 | 0.006 | 0.003 | 0.007 | 0.005 | 0.004 | 0.003 | 0.005 | 0.005 |
| Sum cat | 3.000 | 3.000 | 3.004 | 3.003 | 3.009 | 3.004 | 3.003 | 3.002 | 2.995 | 2.997 | 2.995 | 2.997 | 3.000 | 2.997 | 3.000 | 3.003 | 3.005 | 2.997 | 2.996 | 3.002 |

Mg# = 100 x [Mg/(Mg + Fe)]; cr# = 100 x [Cr/(Cr + Al)]; Fe²⁺ and Fe³⁺ calculated by stoichiometry of the formula unit.

Table 6: Trace element contents (ppm) of Estancia Sol de Mayo (ESM) clinopyroxenes.

| Sample phase | MGP2b | | | | | | | | | | MGP2b2 | | | | | | | | | | | | | |
|--------------|-------------|------|------|------|------|-------------|------|------|------|------|-------------|------|------|------|------|-------------|------|------|------|------|------|------|-----|-----|
| | Lherzollite | | | | | Lherzollite | | | | | Lherzollite | | | | | Lherzollite | | | | | | | | |
| | cpx2 | cpx2 | cpx2 | cpx2 | cpx2 | cpx1 | cpx1 | cpx1 | cpx1 | cpx1 | cpx1 | cpx1 | cpx1 | cpx1 | cpx1 | cp2 | cp2 | cp2 | cp2 | cp2 | cp2 | cp2 | cp2 | cp2 |
| Ba | 5.40 | n.d. | n.d. | 1.57 | | 0.65 | 1.39 | 0.33 | n.d. | 1.42 | 0.40 | 1.21 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | 1.74 | n.d. | 0.10 | | |
| Th | 14.2 | n.d. | 10.4 | n.d. | | 3.29 | 2.75 | 2.35 | n.d. | 3.06 | 2.35 | 2.00 | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | n.d. | 6.26 | 0.24 | 4.24 | | |
| Nb | 1.21 | 1.58 | 0 | 1.11 | | 1.15 | 0.80 | 0.7 | 0.63 | 0.76 | 0.66 | 0.55 | 0.52 | 0.70 | 0.93 | 0.70 | 0.90 | 0.90 | 0.94 | 0.53 | 0.67 | | | |
| La | 3.84 | 4.69 | 4.14 | 4.25 | | 3.77 | 3.70 | 3.42 | 3.74 | 4.28 | 3.64 | 3.61 | 3.95 | 3.69 | 3.02 | 4.24 | n.d. | n.d. | n.d. | n.d. | 5.41 | 9.38 | | |
| Ce | 9.10 | 11.3 | 10.5 | 10.8 | | 9.86 | 9.89 | 9.10 | 9.56 | 8.45 | 9.81 | 8.01 | 9.17 | 8.22 | 8.58 | n.d. | n.d. | n.d. | n.d. | n.d. | 14.1 | 26.7 | | |
| Sr | 97.4 | n.d. | n.d. | n.d. | | 96.9 | 102 | 98.3 | 102 | 101 | 98.6 | 90.1 | 102 | 95.7 | 92.8 | 124 | 123 | 108 | 128 | 201 | | | | |
| Nd | 5.66 | 7.43 | 6.35 | 5.95 | | 4.93 | 5.89 | 4.95 | 5.37 | 4.45 | 6.21 | n.d. | 5.84 | 4.38 | n.d. | 9.05 | 9.32 | n.d. | n.d. | 8.20 | 14.8 | | | |
| Zr | 21.3 | 21.5 | 19.6 | 18.0 | | 21.7 | 20.7 | 18.4 | 16.4 | 19.1 | 17.1 | 8.79 | 12.6 | 19.4 | 19 | 31.4 | 28.6 | 27.0 | 19.4 | 74.2 | | | | |
| Sm | 1.49 | 2.34 | 2.12 | 2.28 | | 1.54 | 1.82 | 1.25 | 0.66 | 1.31 | 1.57 | n.d. | 0.92 | 1.73 | 0.93 | 3.08 | n.d. | n.d. | 2.22 | 1.84 | 3.31 | | | |
| Eu | 0.45 | 0.41 | 0.64 | 0.74 | | 0.37 | 0.57 | 0.59 | 0.32 | 0.46 | 0.32 | 0.51 | 0.21 | 0.38 | 0.28 | n.d. | n.d. | n.d. | 0.87 | 0.51 | n.d. | | | |
| Ti | 1694 | 1565 | 1452 | 1432 | | 1726 | 1502 | 1330 | 1276 | 1479 | 1275 | 1105 | 1092 | 1455 | 1479 | 1290 | 1295 | 1285 | 1119 | 2622 | | | | |
| Gd | 1.60 | n.d. | 1.67 | n.d. | | 1.19 | 1.60 | 0.72 | 1.48 | n.d. | n.d. | 1.60 | n.d. | 1.34 | n.d. | 1.66 | 2.88 | 2.85 | n.d. | n.d. | | | | |
| Dy | 1.99 | 1.43 | 1.70 | 0.99 | | 1.62 | 1.92 | 1.40 | 1.80 | 1.97 | 1.81 | 1.71 | n.d. | 1.24 | n.d. | 2.13 | 2.39 | 2.56 | n.d. | 2.30 | | | | |
| Er | 1.40 | 0.79 | 1.36 | 0.78 | | 1.04 | n.d. | 0.83 | 0.39 | 1.44 | 1.24 | n.d. | 0.98 | 1.32 | 0.79 | 1.03 | 1.46 | 1.52 | 1.17 | 1.11 | | | | |
| Yb | 1.54 | 1.01 | 1.42 | n.d. | | n.d. | 1.25 | 1.25 | 0.71 | n.d. | 0.95 | n.d. | 0.62 | 1.20 | 1.40 | 1.03 | 1.62 | 1.39 | 1.05 | 1.33 | | | | |
| Lu | 0.26 | 0.18 | n.d. | n.d. | | n.d. | 0.18 | 0.17 | n.d. | n.d. | n.d. | 0.14 | n.d. | 0.22 | n.d. | n.d. | 0.23 | 0.27 | 0.13 | n.d. | | | | |

n.d.: not detected.

Table 7: Trace element contents (ppm) of Estancia Sol de Mayo (ESM) orthopyroxenes.

| Sample Rock type | MGP2b | | MGP2b2 | | MGP1b | | MGP1c | | MGP1g | | MGP3b | | MGP4b | | MGP1h | | MGP2a | |
|---------------------|--------------------|--------------------|--------------------|--------------------|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|----------------|----------------|----------------|----------------|
| | opx1 Lherzolite | opx1 Lherzolite | opx1 Lherzolite | opx1 Lherzolite | opx1 Harzburgite | opx2 Harzburgite | opx1 Harzburgite | opx2 Harzburgite | opx1 Harzburgite | opx2 Harzburgite | opx1 Harzburgite | opx2 Harzburgite | opx1 Harzburgite | opx2 Harzburgite | opx1 Dumite | opx2 Dumite | opx3 Dumite | opx3 Dumite |
| Ba | 0.28 | 0.65 | 0.61 | 0.38 | 0.61 | n.d. | 0.94 | 0.72 | n.d. | 0.53 | 0.65 | 0.72 | 0.80 | 0.69 | 0.62 | n.d. | n.d. | 0.46 |
| Th | 0.19 | 0.12 | 0.13 | n.d. | 0.13 | 0.17 | 0.11 | 0.11 | 0.10 | 0.19 | n.d. | 0.16 | 0.16 | n.d. | 0.13 | n.d. | n.d. | 0.16 |
| Nb | 0.45 | 0.11 | 0.45 | n.d. | 0.27 | 0.45 | 0.32 | 0.31 | n.d. | 0.46 | n.d. | 0.44 | 0.35 | 0.15 | 0.34 | 0.42 | 0.12 | 0.36 |
| La | 0.12 | 0.06 | 0.35 | n.d. | 0.19 | 0.21 | 0.14 | 0.18 | 0.14 | 0.18 | n.d. | 0.16 | 0.35 | 0.27 | 0.40 | 0.15 | 0.25 | 0.34 |
| Ce | 0.15 | 0.22 | 0.35 | 0.28 | 0.23 | 0.25 | 0.12 | 0.24 | 0.45 | 0.47 | 0.55 | 0.37 | 0.45 | 0.27 | 0.38 | 0.28 | 0.35 | 0.50 |
| Sr | 0.19 | 0.59 | 0.17 | 0.33 | 0.46 | 0.19 | 0.61 | 0.66 | 0.36 | 0.52 | 0.57 | 0.32 | 0.46 | 0.65 | 0.36 | 0.58 | 0.29 | 0.81 |
| Nd | n.d. | 0.62 | 1.35 | 1.25 | 0.45 | 0.61 | n.d. | 1.26 | 0.70 | 0.72 | 0.80 | 0.40 | 0.61 | n.d. | 0.35 | 0.46 | 0.39 | 0.59 |
| Zr | 1.30 | 0.93 | 1.13 | 0.94 | 1.28 | 0.97 | 2.01 | 2.26 | 5.54 | 4.74 | 5.04 | 1.02 | 1.33 | 1.52 | 2.43 | 1.27 | 2.64 | 20.0 |
| Sm | 0.32 | 0.35 | 0.63 | 0.59 | 0.31 | 0.45 | 0.61 | 0.68 | 0.35 | 0.34 | 0.40 | 0.33 | 0.21 | 0.17 | 0.18 | 0.28 | 0.22 | 0.37 |
| Eu | 0.19 | 0.18 | 0.29 | 0.25 | 0.23 | n.d. | 0.29 | n.d. | 0.17 | n.d. | 0.21 | 0.18 | 0.11 | n.d. | 0.12 | n.d. | 0.23 | 0.21 |
| Ti | 408 | 508 | 490 | 519 | 357 | 300 | 248 | 358 | 1306 | 1286 | 1081 | 743 | 331 | 314 | 435 | 254 | 217 | 635 |
| Gd | 0.81 | 0.82 | n.d. | 1.00 | 1.09 | 1.05 | 1.21 | 1.25 | 0.65 | 0.75 | 0.86 | 0.79 | 0.50 | 0.40 | 0.47 | 0.53 | 0.45 | 0.81 |
| Dy | 1.06 | 1.09 | 1.31 | 1.07 | 1.37 | 1.32 | 1.23 | 1.28 | 0.85 | 1.00 | 1.20 | 1.08 | 0.56 | 0.49 | 0.60 | 0.67 | 0.58 | 1.02 |
| Er | 0.69 | 0.71 | 0.84 | 0.69 | 0.89 | 0.86 | 0.75 | 0.74 | 0.60 | 0.71 | 0.81 | 0.78 | 0.36 | 0.33 | 0.39 | 0.41 | 0.36 | 0.67 |
| Yb | 0.70 | 0.72 | 0.82 | 0.73 | 0.90 | 0.88 | 0.74 | 0.72 | 0.61 | 0.72 | 0.81 | 0.77 | 0.35 | 0.33 | 0.39 | 0.39 | 0.36 | 0.68 |
| Lu | n.d. | n.d. | 0.12 | 0.10 | 0.14 | 0.13 | 0.12 | 0.12 | n.d. | 0.11 | n.d. | 0.12 | 0.06 | 0.05 | 0.06 | n.d. | 0.06 | 0.10 |

n.d.: not detected.

Table 8: Equilibration temperature, pressure and fO_2 estimates of Estancia Sol de Mayo (ESM) mantle xenoliths.

| Sample | Lithology | Area | Cpx | Opx | Ol | T°C ⁺ | mean T°C | T°C* | P Kbar | $\Delta \log fO_2$ |
|--------|--------------|--------|-----|-----|----|------------------|-------------|------------|--------|--------------------|
| MGP2b | Lherzolitite | Area 2 | 7 | 6 | 4 | 1055 | | | 33 | |
| | | Area 6 | 3 | 5 | 8 | 1011 | | | 20 | |
| | | Area 8 | 6 | 11 | 8 | 1026 | 1031 | 945 | 38 | 0.02 |
| MGP2b2 | Lherzolitite | Area 2 | 8 | 9 | 5 | 1026 | | | 16 | |
| | | Area 1 | 3 | 4 | 8 | 1041 | | | 18 | |
| | | Area 7 | 10 | 5 | 5 | 943 | 1003 | 913 | 3 | 0.62 |
| MGP1b | Harzburgite | Area 2 | 2 | 1 | 3 | 1041 | | | 24 | |
| | | Area 3 | 1 | 2 | 3 | 1008 | | | 2 | |
| | | Area 5 | 2 | 1 | 4 | 995 | | | -5 | |
| | | Area 6 | 4 | 3 | 3 | 992 | | | -2 | |
| | | Area 7 | 3 | 2 | 5 | 1002 | | | 1 | |
| | | Area 8 | 3 | 4 | 6 | 1017 | 1009 | 912 | 3 | 1.26 |
| MGP3b | Harzburgite | Area 4 | 2 | 3 | 9 | 1012 | | | 15 | |
| | | Area 4 | 8 | 6 | 9 | 1017 | | | 16 | |
| | | Area 5 | 2 | 4 | 5 | 1003 | | | 5 | |
| | | Area 7 | 2 | 8 | 3 | 1028 | 1015 | 954 | 12 | 1.47 |
| MGP4b | Harzburgite | Area 3 | 1 | 6 | 1 | 1044 | | | 13 | |
| | | Area 4 | 7 | 9 | 2 | 1027 | | | 11 | |
| | | Area 5 | 1 | 6 | 6 | 1037 | | | 2 | |
| | | Area 7 | 10 | 14 | 8 | 1040 | | | 13 | |
| | | Area 7 | 3 | 6 | 15 | 1053 | 1040 | 980 | -10 | 1.01 |

T°C⁺ calculated after Brey and Köhler (1990)

T°C* calculated after Ballhaus et al. (1991)

P Kbar calculated after Köhler and Brey (1990)

 $\Delta \log fO_2$ calculated using T after Brey and Köhler (1990)*italics* unreasonably low and/or high and negative pressures

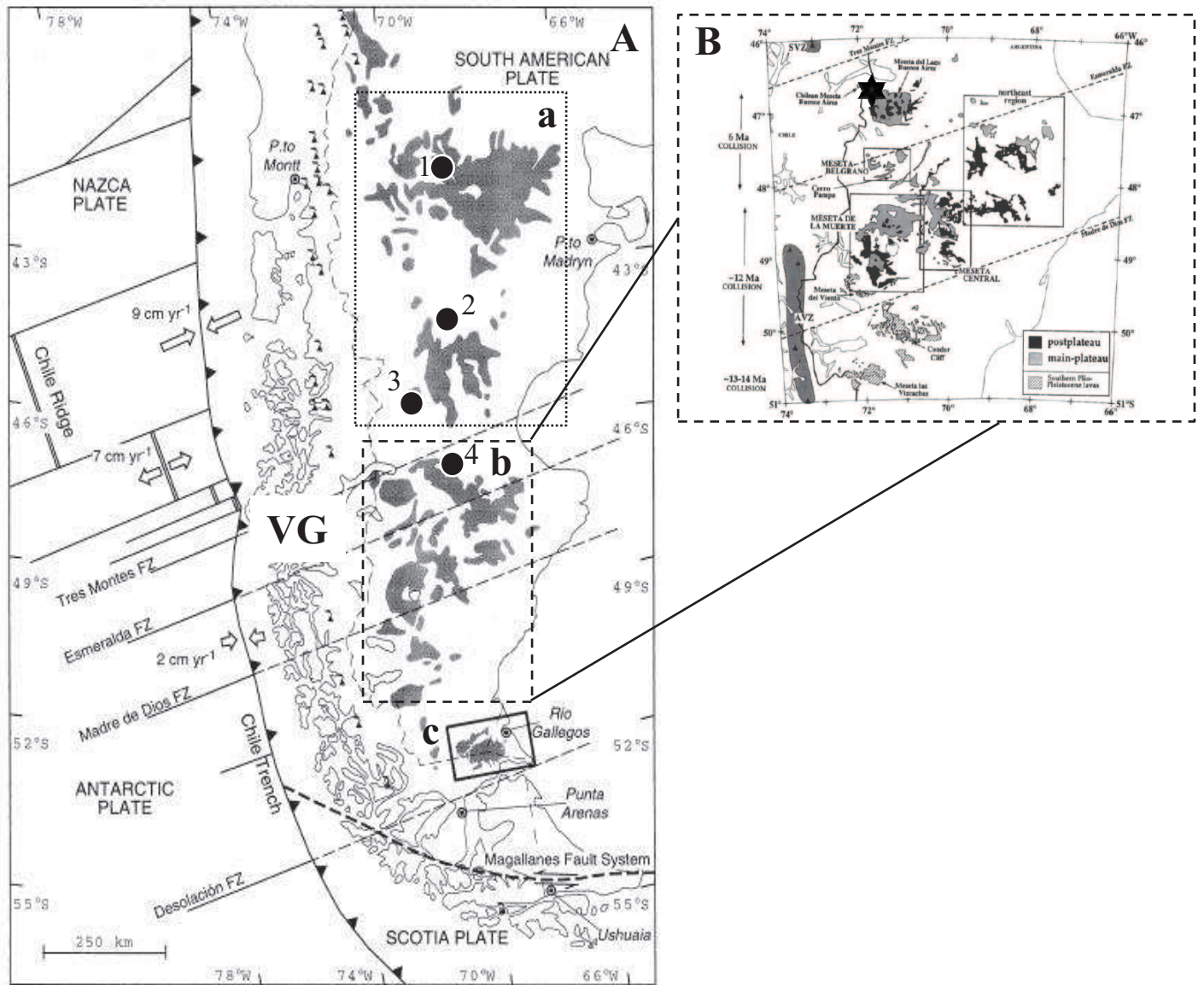


Fig. 1: Sketch map of Patagonia (A, after *D'Orazio et al., 2000*). VG stands for “Volcanic Gap”. (a), (b) and (c) indicate the back-arc volcanic fields respectively of Northern Patagonia, Central Patagonia and Southern Patagonia. 1 Cerro Aznare; 2 Cerro Rio Chubut; 3 Cerro de los Chenques; 4 Cerro Clark. Sketch map B (from *Gorring et al., 1997*) shows the occurrence of the different plateau of Central Patagonia. In grey and black are represented the main and post-plateau sequences respectively. Black star localizes sampling site of xenoliths at Estancia Sol de Mayo (ESM).

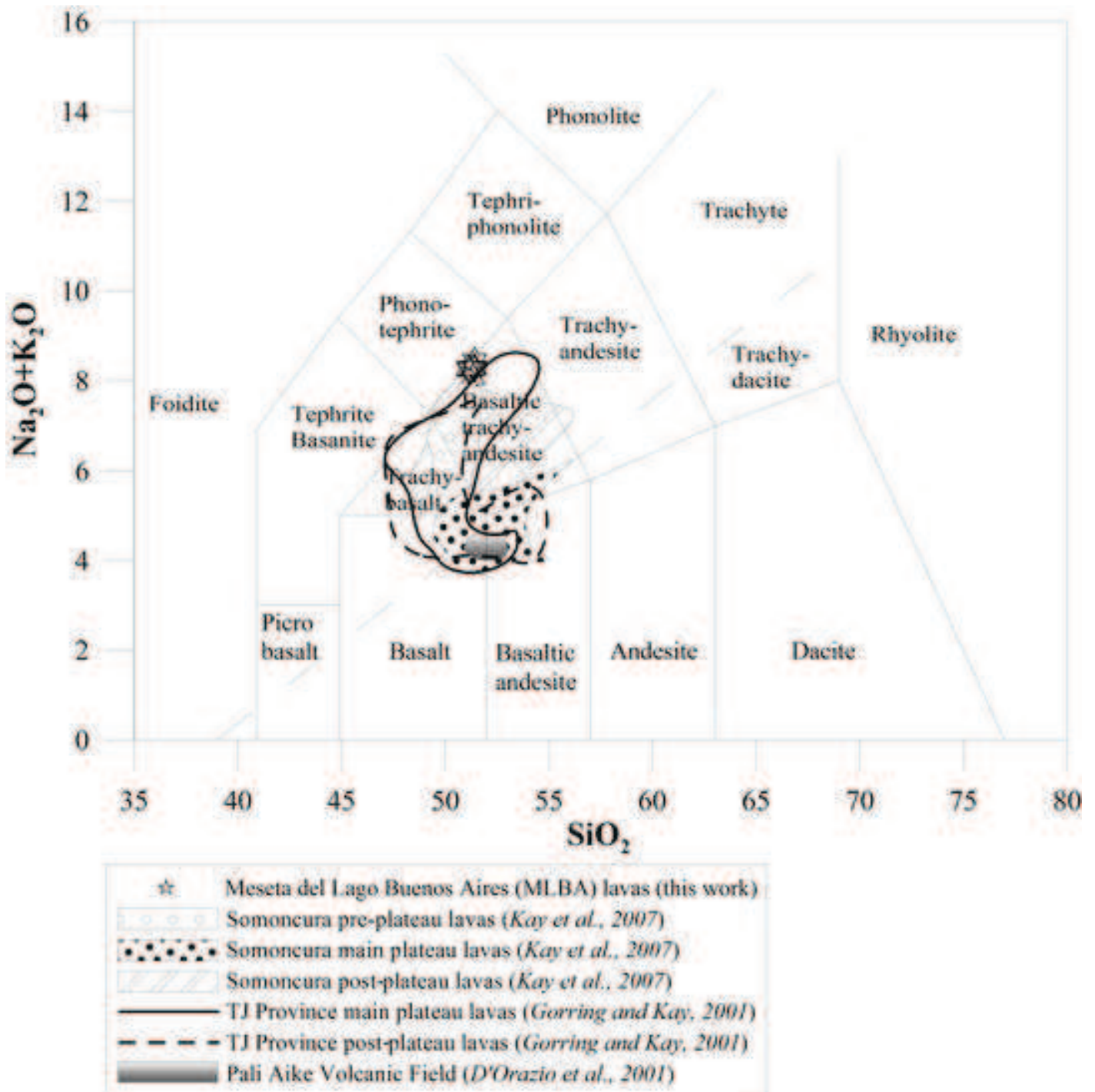


Fig. 2: Total alkali vs. silica diagram of Le Bas and Streckeisen (1991). Dash dot line separates the alkaline and subalkaline domains.

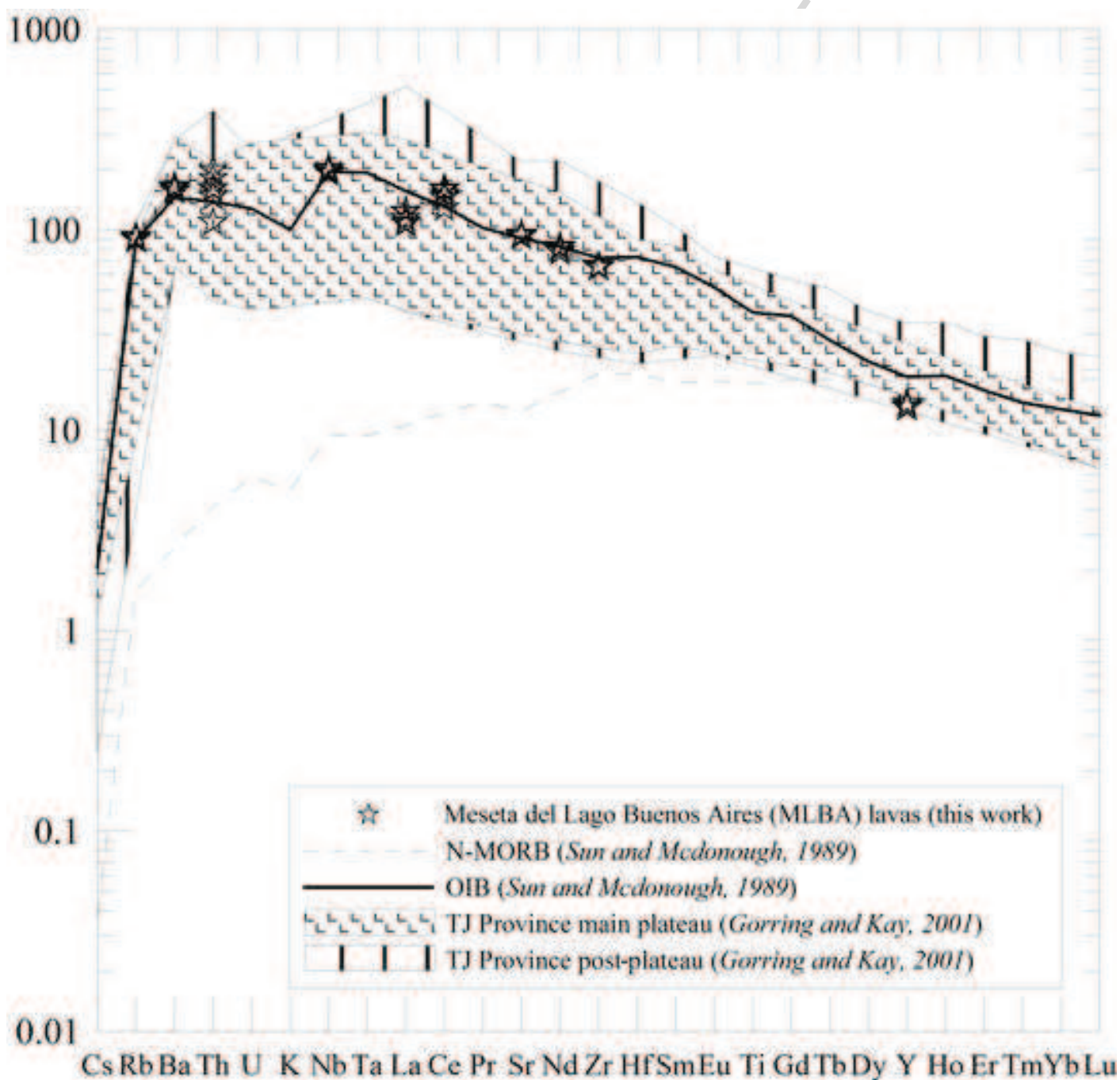


Fig. 3: Chondrite normalized (Sun and McDonough, 1989) trace element compositions of Meseta Lago Buenos Aires (MLBA) post-plateau lavas.

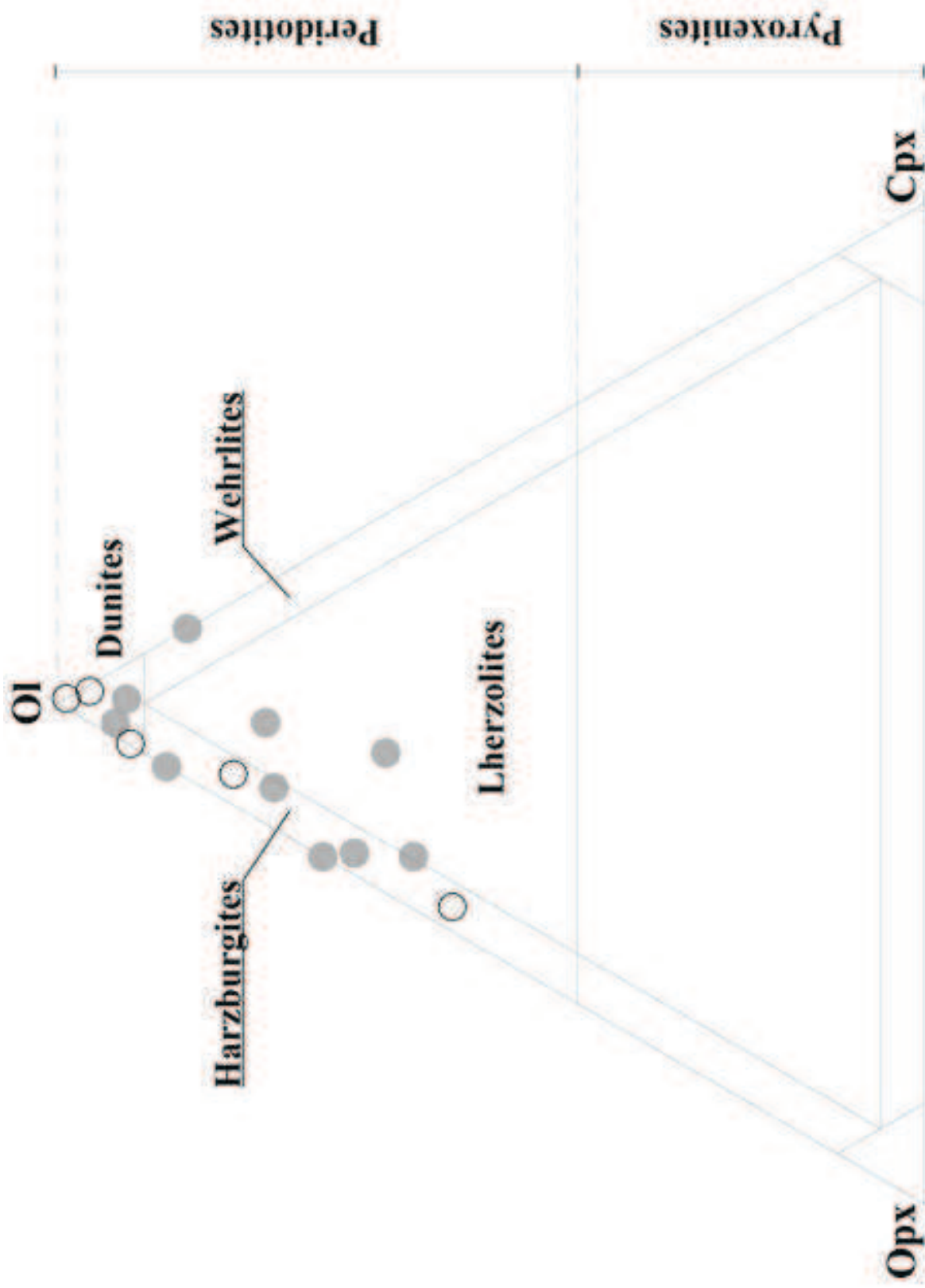


Fig. 4: Ultramafic classification diagram (after Streckeisen, 1976) of the Estancia Sol de Mayo (ESM) mantle xenoliths. Empty symbols indicate samples studied only petrographically, while full symbols indicate those studied both petrographically and geochemically.

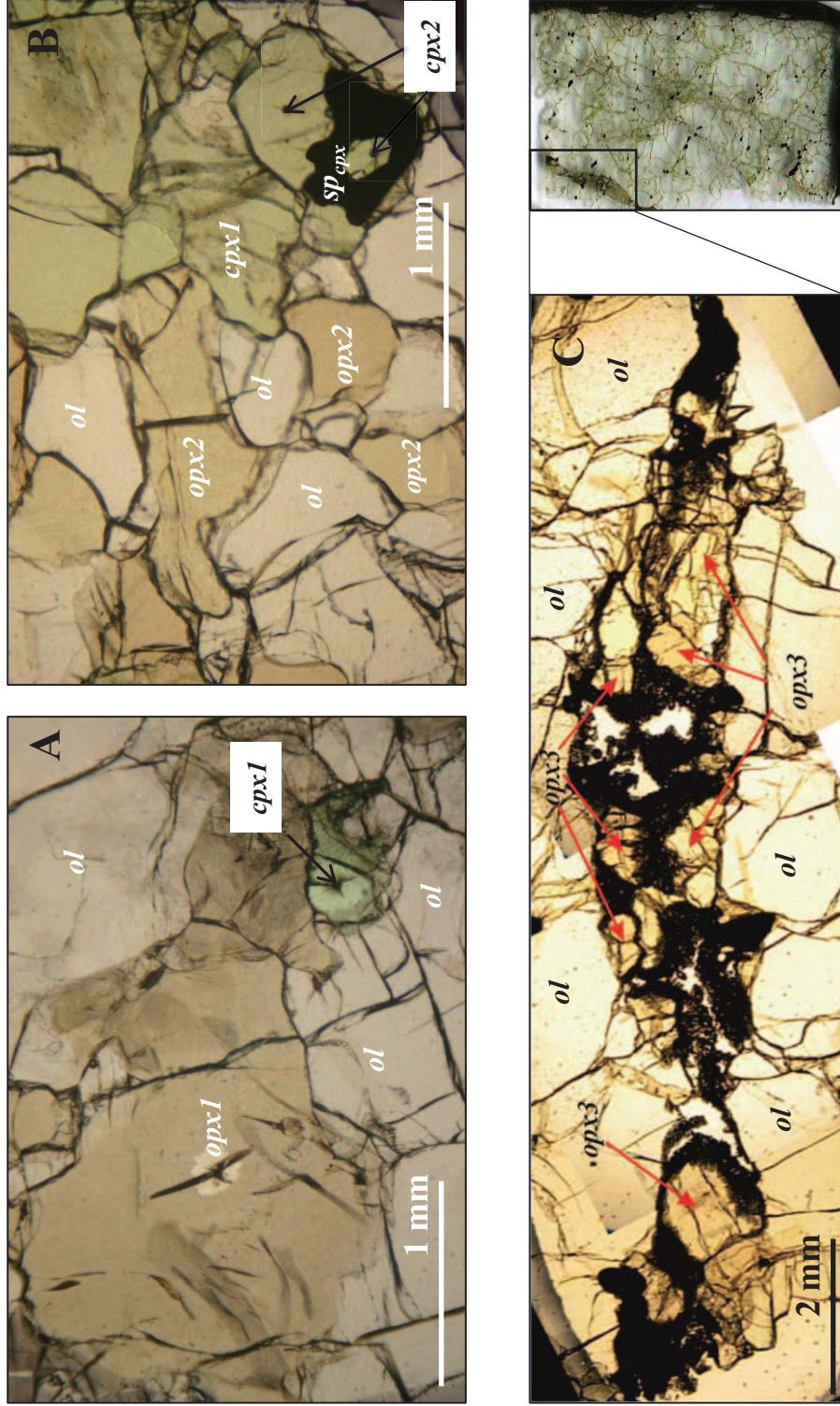


Fig. 5: Transmitted plane-polarized photomicrographs of representative assemblages in the Estancia Sol de Mayo (ESM) xenoliths. Ol, olivine; opx, orthopyroxene; cpx, clinopyroxene; sp, spinel. Cpx are further classified as cpx1 and cpx2. The former generally occur as protogranular in the peridotitic matrix, while the latter is observed around the sp. Opx is subdivided in opx1, opx2 and opx3: the first present as large protogranular crystals with exsolution lamellae while the second as small clean and undeformed grains without exsolution lamellae; the third occur as smaller grains arranged in vein. (A) Protogranular anhydrous spinel-bearing harzburgite MGP4b comprising ol, cpx1 and opx1. (B) Protogranular to porphyroclastic anhydrous spinel-bearing harzburgite MGP1b comprising ol, small porphyroclastic clean and undeformed grains of opx2, a cpx1 grain and a cpx2 growing around a black sp. (C) Vein of opx3 in dunite MGP2a. Opx3 are surrounded by a black matrix constituted by very small grains of ol, cpx and plagioclase.

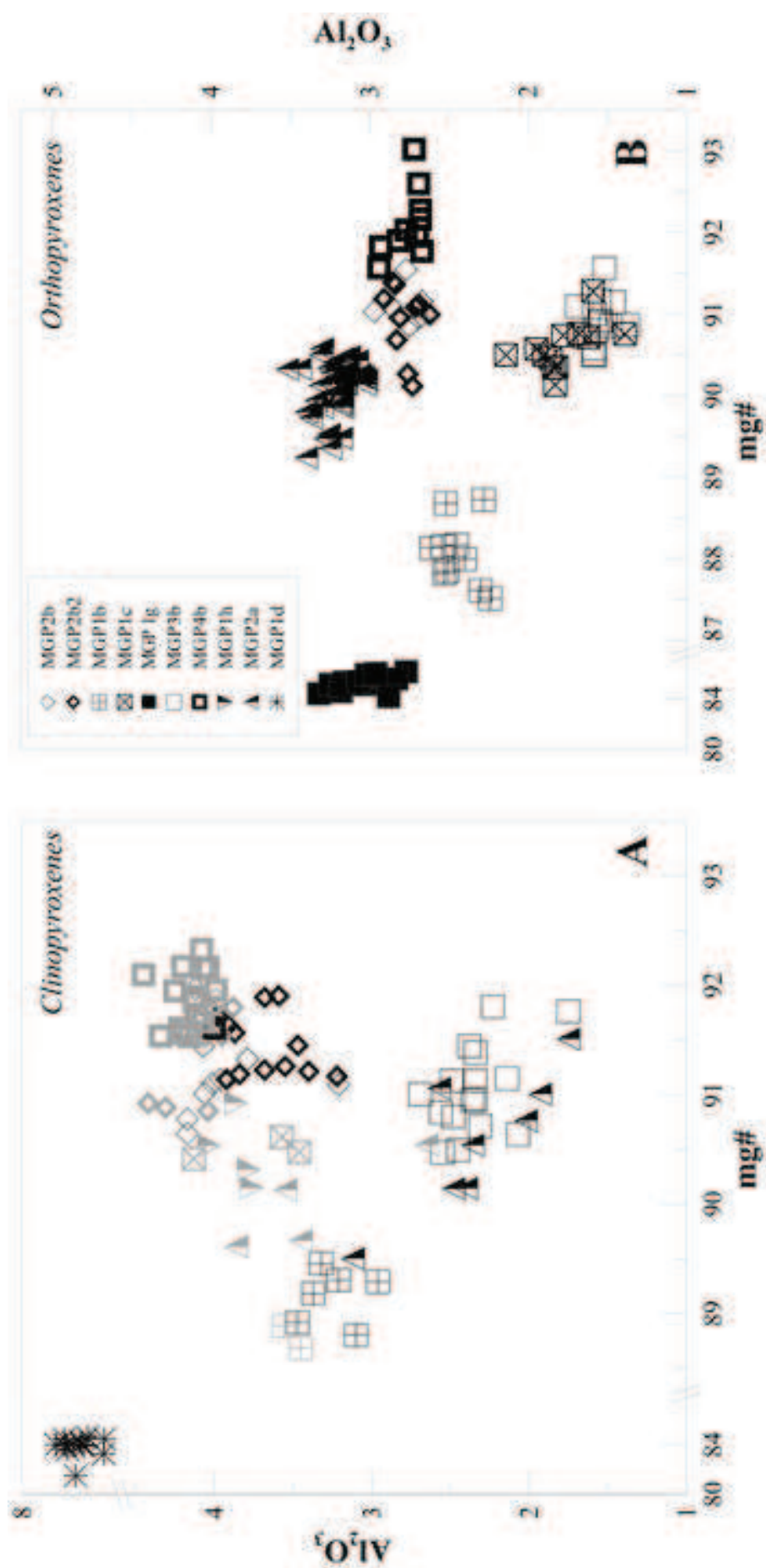


Fig. 6: Al_2O_3 vs $\text{mg}\#$ of clinopyroxenes and orthopyroxenes. Diamonds refer to ilherzschites, squares to harzburgites, triangles to dimites and asterisk to wehrlite. In A, black symbols represent epx1 while grey symbols epx2 .

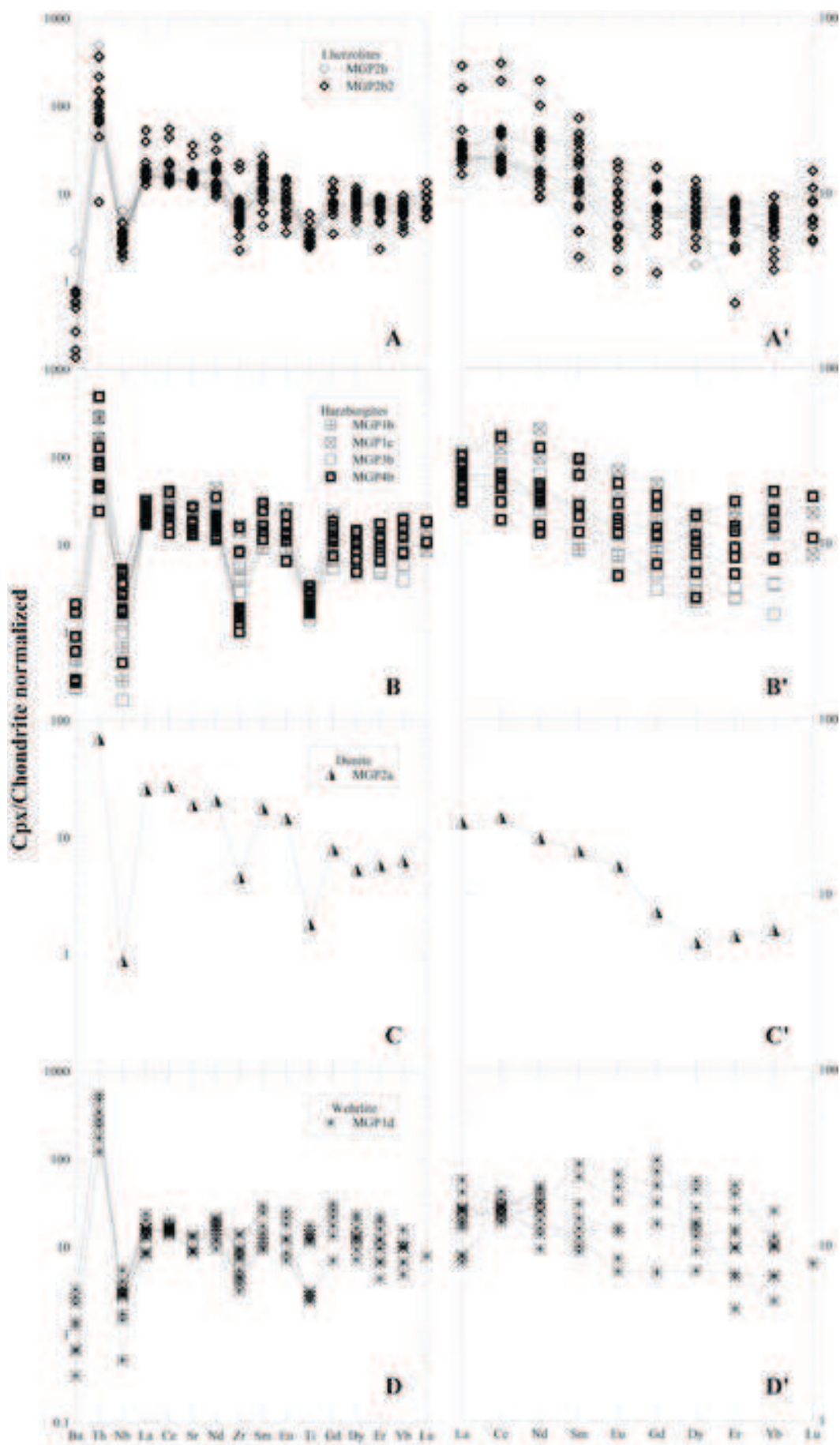


Fig. 3. Chondrite-normalized REE and Mg-bearing mineral patterns for (A, B, C, D) and REF (A', B', C', D') of chondrites.

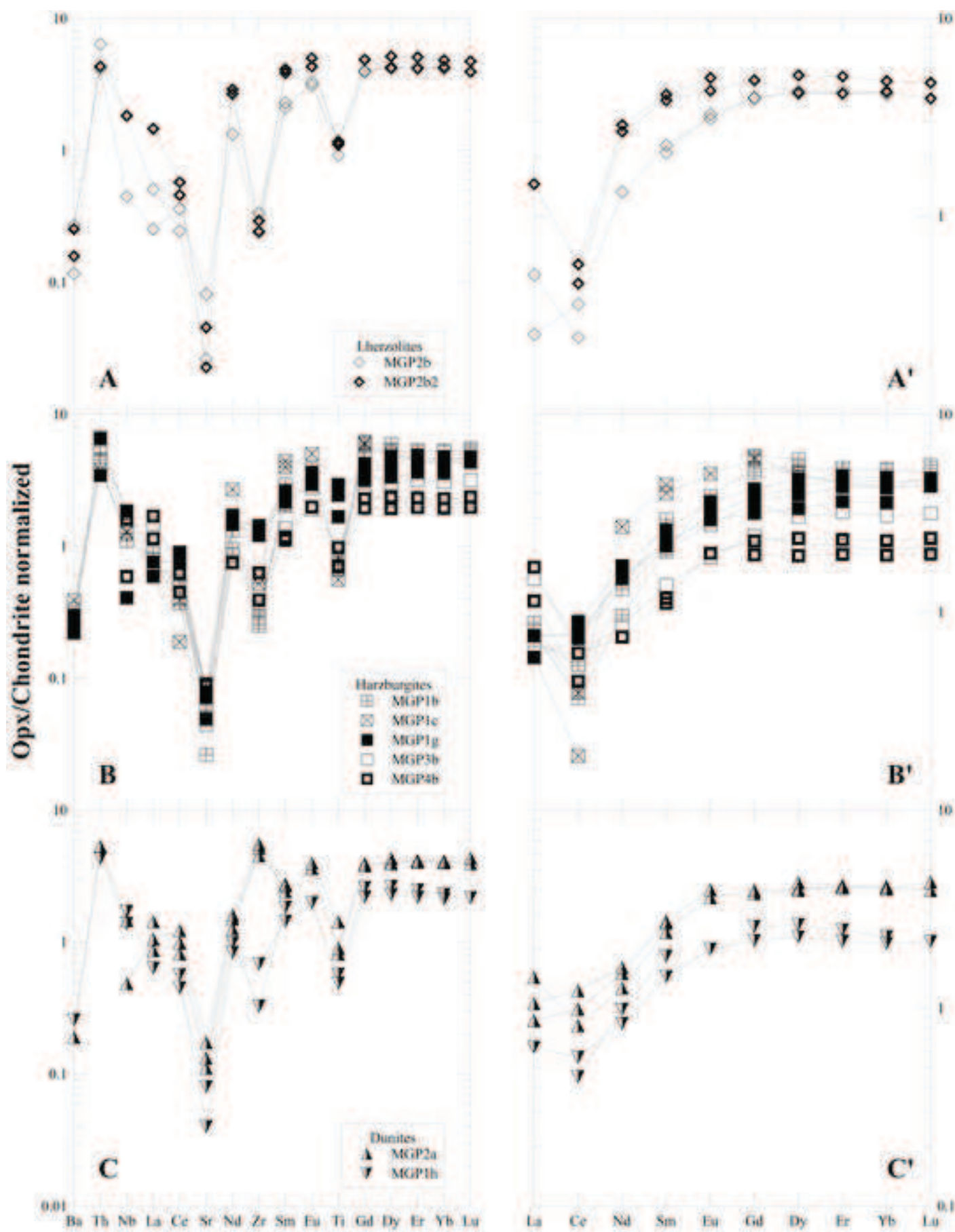


Fig. 8) Chondrite-normalized (Sun and McDonough, 1989) incompatible trace elements (A, B, C) and REE (A', B', C') of orthopyroxenes.

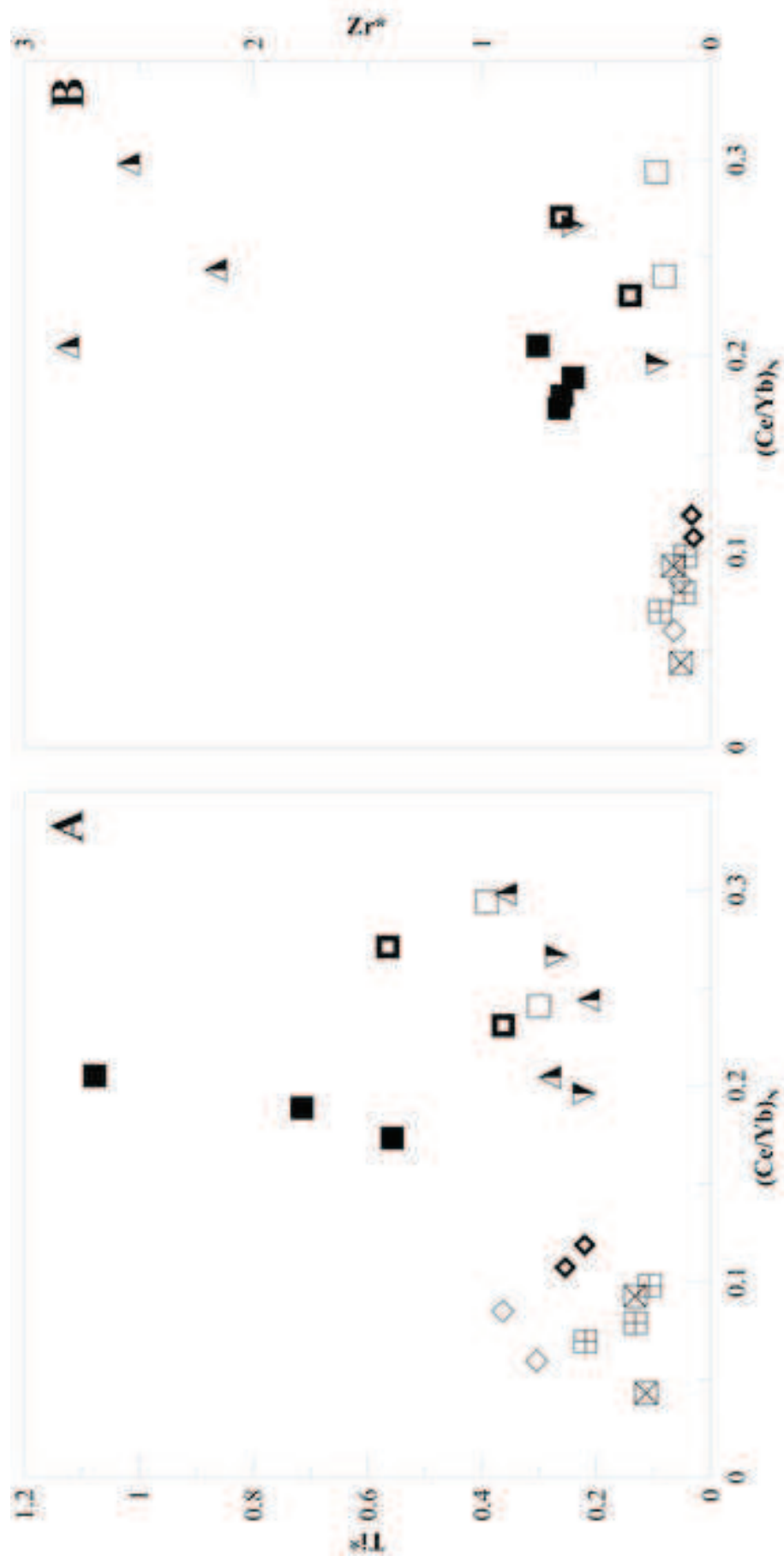


Fig. 9: Ti^* (calculated as $[Ti_N / ((Ba_N + Cd_N) / 2)]$) and Zr^* (calculated as $[Zr_N / ((Sm_N + Nd_N) / 2)]$) vs. $(Ce/Yb)_N$ for some selected orthopyroxenes from Estancia Sol de Mayo (ESM). For symbols refer to Fig. 6.

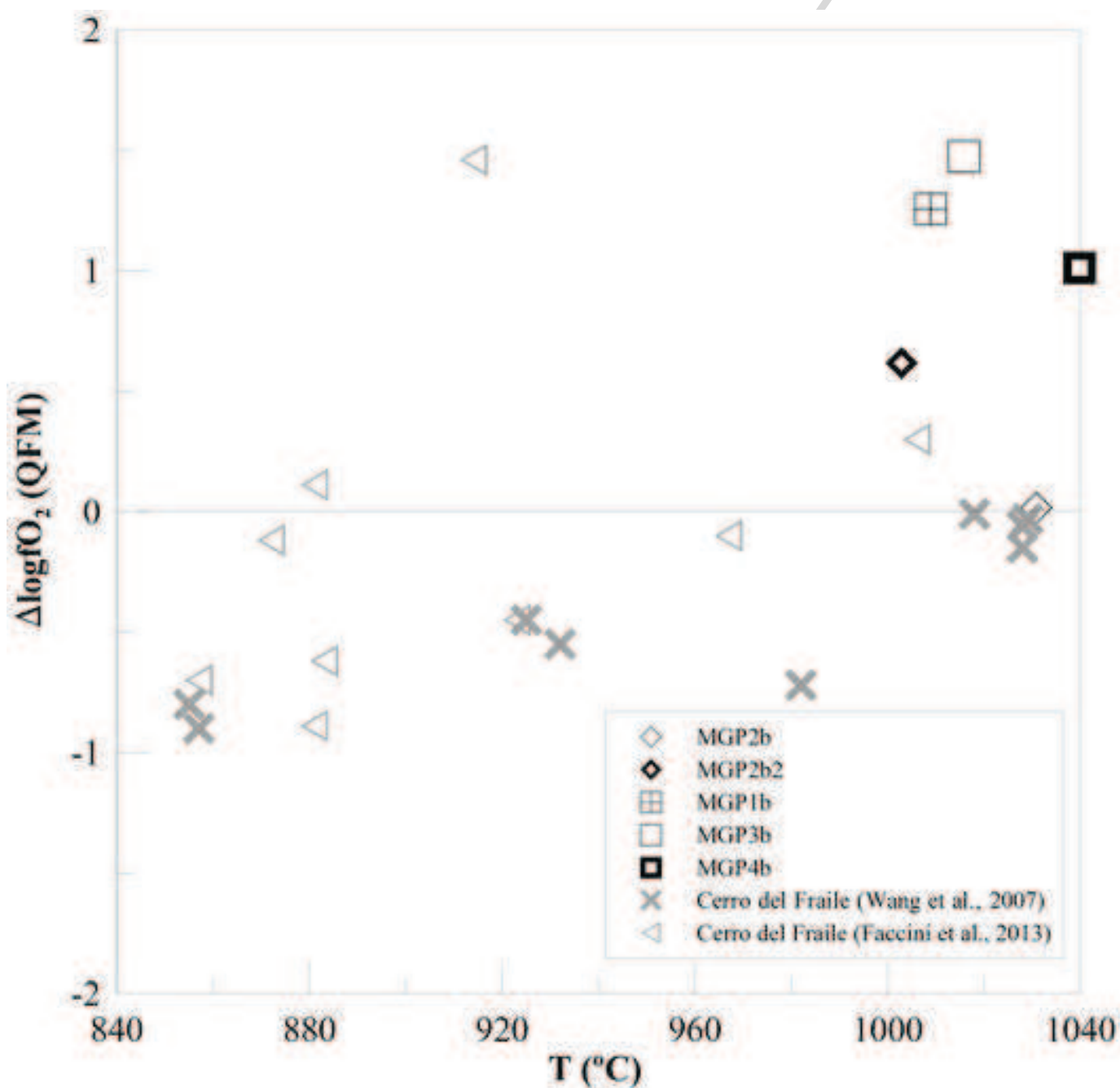


Fig. 10: Oxygen fugacity [calculated as $\Delta \log f_{O_2}$ (QFM) (Ballhaus et al., 1991)] vs. temperature (from Brey and Köhler, 1990) of Estancia Sol de Mayo (ESM) peridotites. For comparison oxygen fugacity of Cerro del Fraile from Wang et al. (2007) and Faccini et al. (2013).

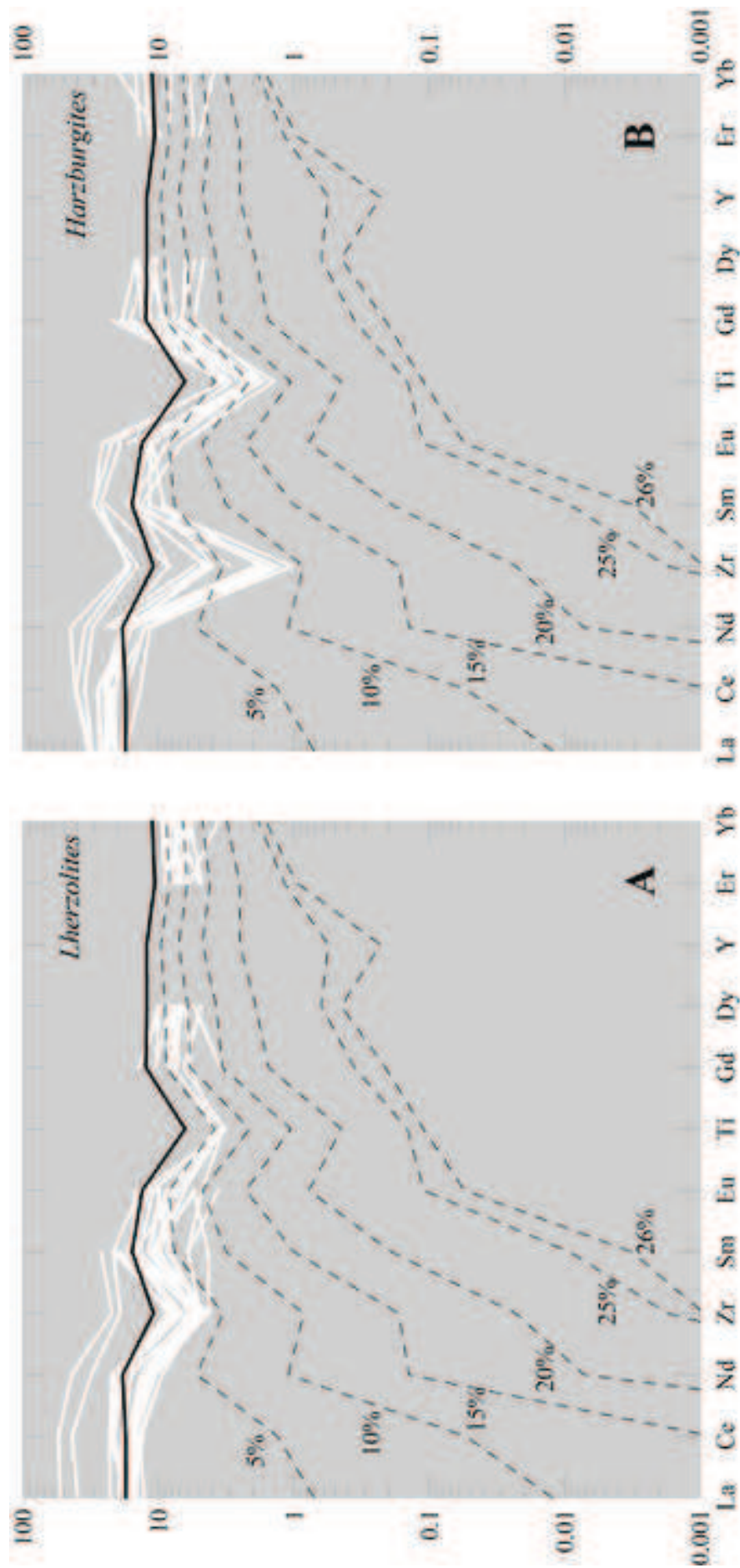


Fig. 11: Chondrite-normalized REE of epx from Estancia Sol de Mayo (ESM, white lines) from ilmenites (A) and harzburgites (B) compared to the curves (black dashed lines) of 5%, 10%, 15%, 20%, 25% and 26% fractional partial melting (Johnson et al., 1990) of a starting fertile epx (bold black line) from Boesdirtman et al. (2005).

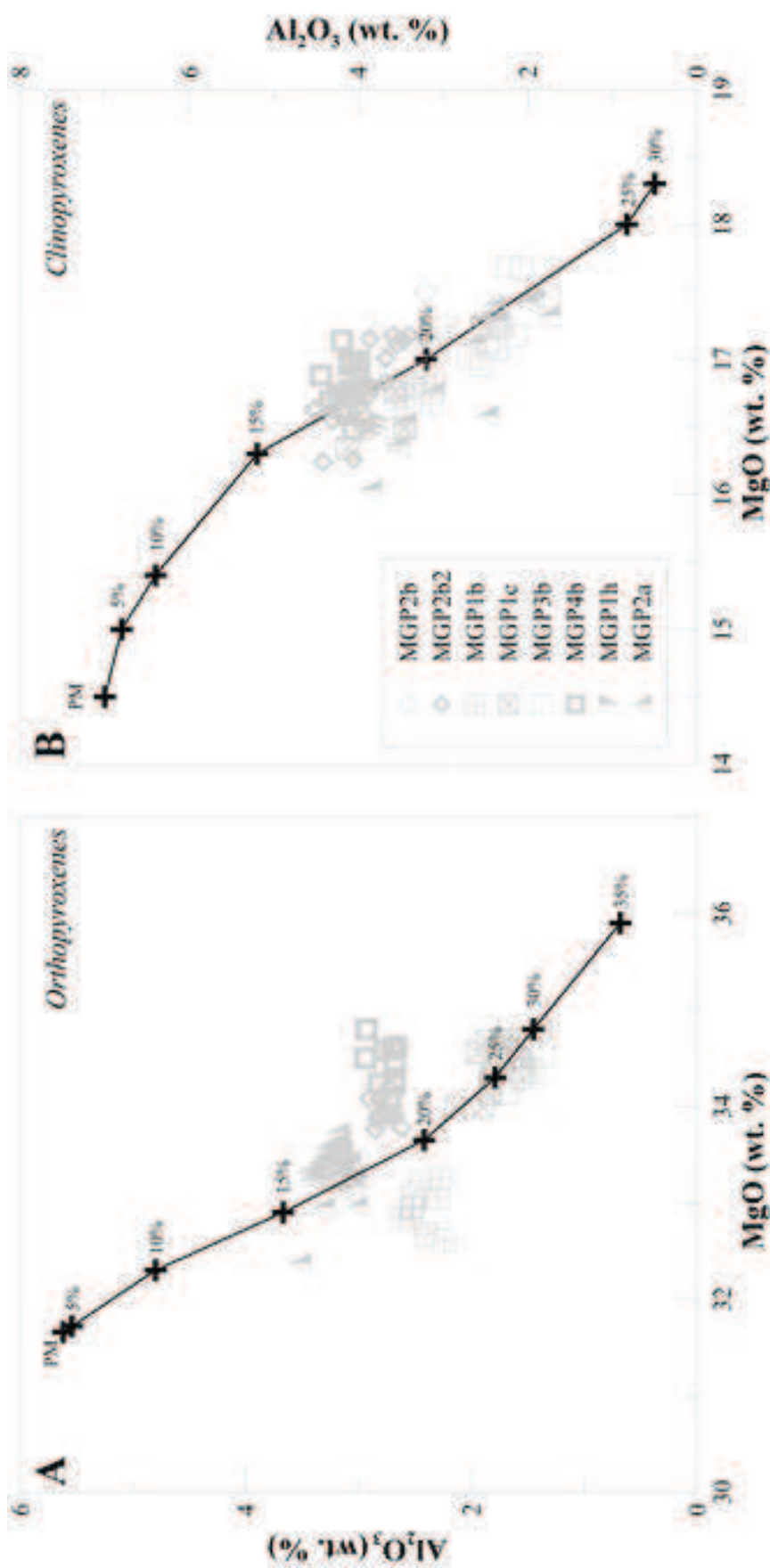


Fig. 12: Al₂O₃ vs MgO melting trends from Botcharnikov et al., (2011) for opx and cpx in Estancia Sol de Mayo (ESM) mantle xenoliths. Primitive Mantle (PM) opx and cpx composition in terms of Al₂O₃ and MgO were calculated on the basis of the primitive mantle composition of McDonough & Sun (1995). Black crosses on curves indicate partial melting percentages.

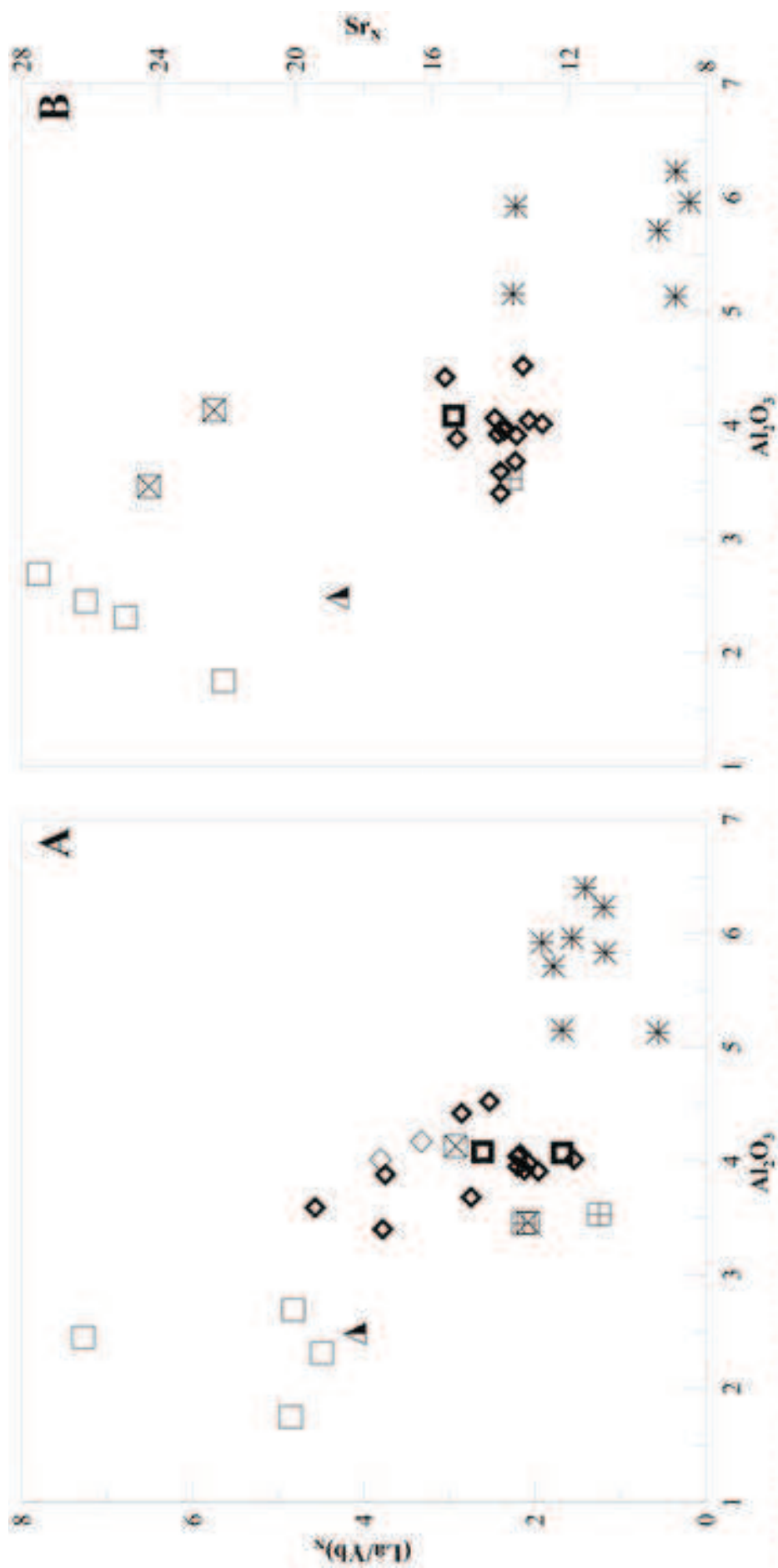


Fig. 13: Variation in some selected samples of (La/Yb)_N and Sr_N vs. Al₂O₃ of clinopyroxenes. A negative correlation between the two geochemical markers and the content of Al₂O₃ is highlighted. For symbols refer to Fig. 6.

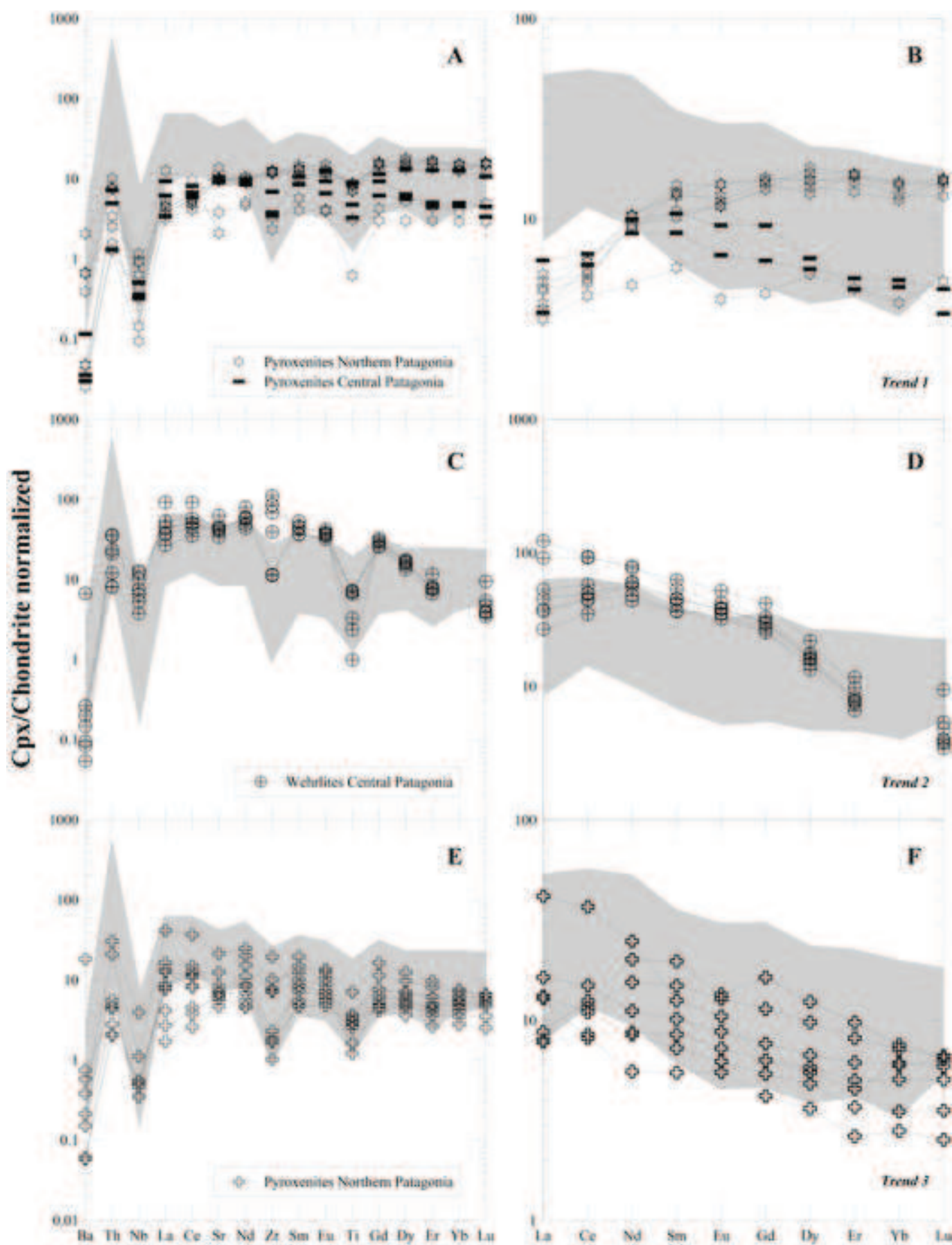


Fig. 14. Chondrite-normalized (Sun and McDonough, 1989) incompatible trace elements (A, C, E) and REE patterns (B, D, F) of clinopyroxenes from pyroxenites of Northern and Central Patagonia and wehrfites from Central Patagonia (Dumas, 2007). Light gray field represents the clinopyroxenes from Estancia Sol de Mayo (ESM).

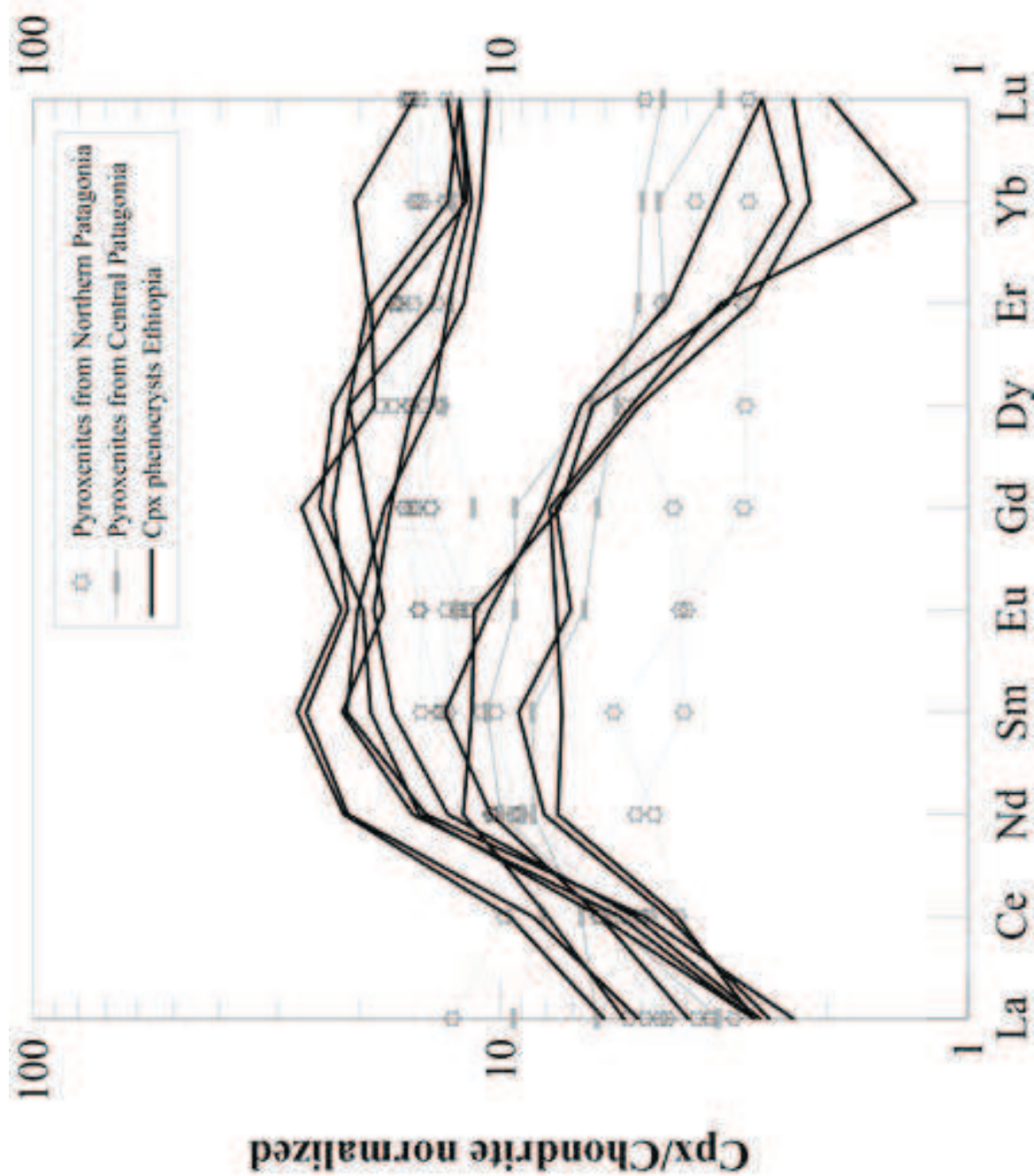


Fig. 15: Chondrite-normalized (Sun and McDonough, 1989) incompatible trace elements of clinopyroxenes from pyroxenites of Northern and Central Patagonia (Dantas, 2007) and clinopyroxenes phenocrysts entrained in Northern Ethiopian continental flood basalts (from Beccaluva et al., 2004).

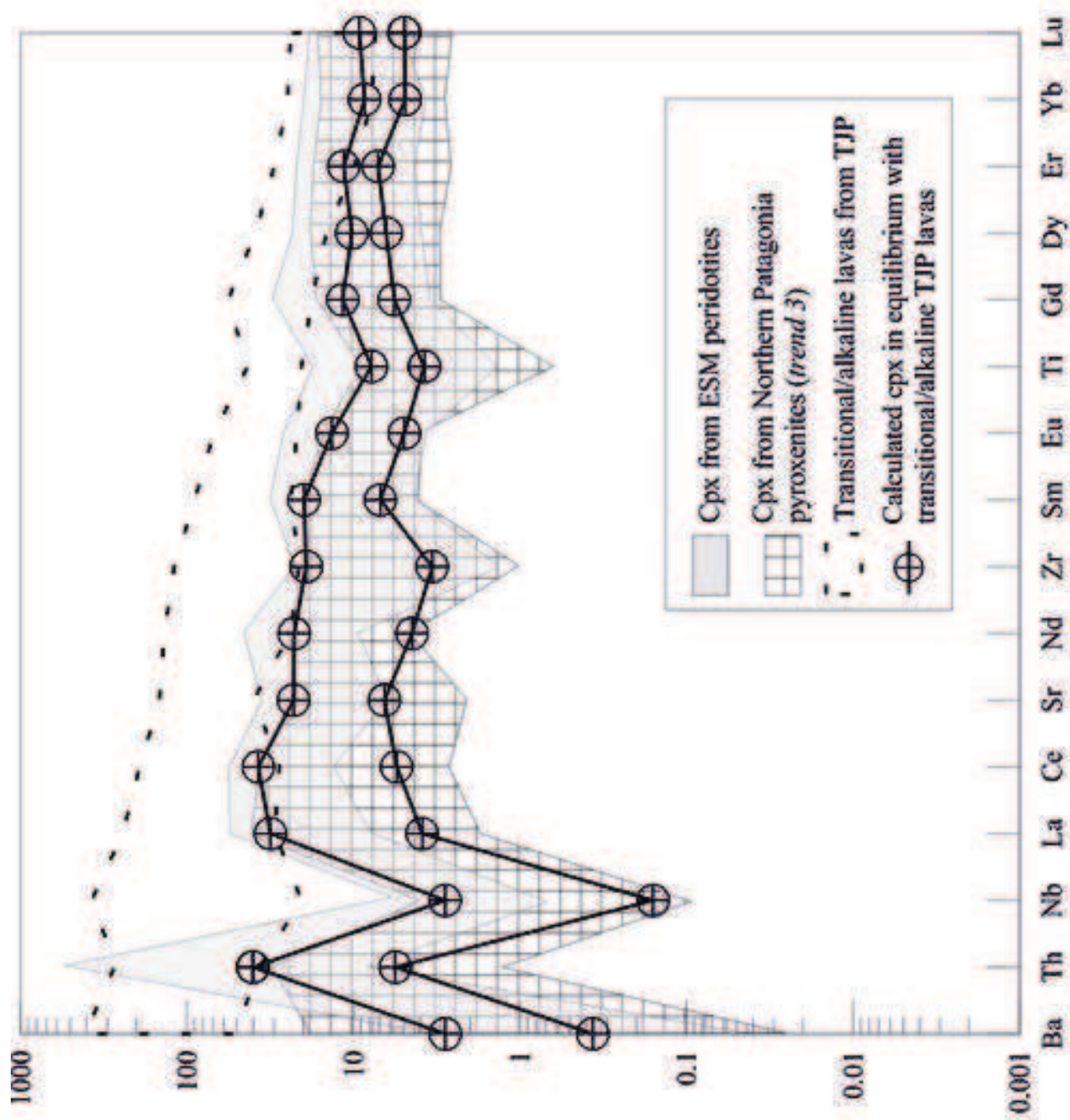


Fig. 16: Chondrite-normalized (Sun and McDonough, 1989) incompatible trace elements pattern reconstructions of clinopyroxenes from transitional/alkaline mafic lavas from the Triple Junction Province.

1 Highlights

2

3 - Two clinopyroxenes and three orthopyroxenes generations in Patagonian peridotites

4 - Al_2O_3 vs. mg# identify two different trends for clinopyroxenes and orthopyroxenes

5 - Clinopyroxenes in ESM peridotites match those of Northern Patagonia pyroxenites

6 - A refertilization process due to a transitional melt occurred at mantle depth

7

8

ACCEPTED MANUSCRIPT